

**PUBLIC HEALTH GOALS FOR
CHEMICALS IN DRINKING WATER**

PERCHLORATE

March 2004

**Governor of the State of California
Arnold Schwarzenegger**

**Secretary for Environmental Protection
California Environmental Protection Agency
Terry Tamminen**

**Director
Office of Environmental Health Hazard Assessment
Joan E. Denton, Ph.D.**



**Public Health Goal for
PERCHLORATE
in Drinking Water**

Prepared by

**Office of Environmental Health Hazard Assessment
California Environmental Protection Agency**

Pesticide and Environmental Toxicology Section

Anna M. Fan, Ph.D., Chief

Robert Howd, Ph.D.

David Ting, Ph.D.

Deputy Director for Scientific Affairs

George V. Alexeeff, Ph.D.

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LIST OF CONTRIBUTORS

PHG PROJECT MANAGEMENT	REPORT PREPARATION	SUPPORT
<i>Project Director</i> Anna Fan, Ph.D.	<i>Author</i> David Ting, Ph.D.	<i>Administrative Support</i> Edna Hernandez Coordinator Sharon Davis Hermelinda Jimenez Genevieve Vivar
<i>Public Workshop</i> Robert Howd, Ph.D. Juliet Rafol	<i>Primary Reviewers</i> Tom McDonald, Ph.D. Martha Sandy, Ph.D. Lauren Zeise, Ph.D.	<i>Library Support</i> Charleen Kubota, M.L.S.
<i>Coordination of External Review</i> Moira Sullivan, M.S.	<i>Final Reviewers</i> Robert Howd, Ph.D. Anna Fan, Ph.D. George Alexeeff, Ph.D.	<i>Web site Posting</i> Edna Hernandez Laurie Monserrat
<i>Revisions/Responses</i> Robert Howd, Ph.D.		
<i>Children's Safety Issues</i> Mark Miller, MD, MPH		

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PREFACE

**Drinking Water Public Health Goals
Pesticide and Environmental Toxicology Section
Office of Environmental Health Hazard Assessment
California Environmental Protection Agency**

This Public Health Goal (PHG) technical support document provides information on health effects from contaminants in drinking water. PHGs are developed for chemical contaminants based on the best available toxicological data in the scientific literature. These documents and the analyses contained in them provide estimates of the levels of contaminants in drinking water that would pose no significant health risk to individuals consuming the water on a daily basis over a lifetime. The PHG is a drinking water goal only; therefore, this document does not evaluate the safe levels of perchlorate in foods or other sources.

The California Safe Drinking Water Act of 1996 (Health and Safety Code, Section 116365) requires the Office of Environmental Health Hazard Assessment (OEHHA) to perform risk assessments and publish PHGs for contaminants in drinking water based exclusively on public health considerations. Section 116365 specifies that the PHG is to be based exclusively on public health considerations without regard to cost impacts. The Act requires that PHGs be set in accordance with the following criteria:

1. PHGs for acutely toxic substances shall be set at levels at which no known or anticipated adverse effects on health will occur, with an adequate margin of safety.
2. PHGs for carcinogens or other substances that can cause chronic disease shall be based upon currently available data and shall be set at levels that OEHHA has determined do not pose any significant risk to health.
3. To the extent the information is available, OEHHA shall consider possible synergistic effects resulting from exposure to two or more contaminants.
4. OEHHA shall consider the existence of groups in the population that are more susceptible to adverse effects of the contaminants than a normal healthy adult.
5. OEHHA shall consider the contaminant exposure and body burden levels that alter physiological function or structure in a manner that may significantly increase the risk of illness.
6. In cases of insufficient data to determine a level of no anticipated risk, OEHHA shall set the PHG at a level that is protective of public health with an adequate margin of safety.
7. In cases where scientific evidence demonstrates that a safe dose-response threshold for a contaminant exists, then the PHG should be set at that threshold.
8. The PHG may be set at zero if necessary to satisfy the requirements listed above.

9. OEHHA shall consider exposure to contaminants in media other than drinking water, including food and air and the resulting body burden.
10. PHGs published by OEHHA shall be reviewed every five years and revised as necessary based on the availability of new scientific data.

PHGs published by OEHHA are for use by the California Department of Health Services (DHS) in establishing primary drinking water standards (State Maximum Contaminant Levels, or MCLs). Whereas PHGs are to be based solely on scientific and public health considerations without regard to economic cost considerations, drinking water standards adopted by DHS are to consider economic factors and technical feasibility. Each standard adopted shall be set at a level that is as close as feasible to the corresponding PHG, placing emphasis on the protection of public health. PHGs established by OEHHA are not regulatory in nature and represent only non-mandatory goals. By federal law, MCLs established by DHS must be at least as stringent as the federal MCL if one exists.

PHG documents are used to provide technical assistance to DHS, and they are also informative reference materials for federal, state and local public health officials and the public. While the PHGs are calculated for single chemicals only, they may, if the information is available, address hazards associated with the interactions of contaminants in mixtures. Further, PHGs are derived for drinking water only and are not intended to be utilized as target levels for the contamination of other environmental media.

The National Academy of Sciences (NAS) is conducting an evaluation of U.S. EPA's 2002 Draft Toxicological and Risk Characterization for Perchlorate. This is an important undertaking that may help guide efforts to study the health effects of perchlorate. When that evaluation is completed, OEHHA will carefully review the NAS conclusions and will revise the PHG as necessary (Health and Safety code section 116365(e)(1)).

Additional information on PHGs can be obtained at the OEHHA Web site at www.oehha.ca.gov.

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PUBLIC HEALTH GOAL FOR PERCHLORATE IN DRINKING WATER

SUMMARY

A Public Health Goal (PHG) of 6 ppb (6 $\mu\text{g/L}$) has been developed for perchlorate in drinking water. The inhibitory effect of perchlorate on the uptake of iodide by the thyroid gland is identified as the critical event for assessing perchlorate risk. Perchlorate competes with iodide at a transport molecule called the "sodium-iodide symporter" (NIS) that is responsible for the active transport of iodide into the thyroid. Iodine is an essential trace element and can be rate limiting in thyroid hormone synthesis. Because perchlorate inhibits or reduces uptake of iodide by the thyroid, many health effects associated with perchlorate exposure are expected to be similar to those caused by iodine deficiency. The nature and severity of the effects are related to the extent and duration of perchlorate exposure as well as the iodine status of the individual. Four sensitive subpopulations are identified in this evaluation: (i) pregnant women and their fetuses, especially those who are getting less than a sufficient amount of iodine; (ii) lactating women, especially those who are getting less than a sufficient amount of iodine, (iii) infants; and (iv) individuals with thyroid problems.

This evaluation included a review of the toxicity data on perchlorate and adverse health effects associated with iodine deficiency. In animal as well as human studies, it has been shown that perchlorate can inhibit the NIS, disrupt thyroid-pituitary homeostasis, and cause pathological changes of the thyroid and sometimes even thyroid tumors. The PHG level of 6 ppb is intended to prevent the very first step, inhibition of NIS. It is believed that by doing so, one can also prevent all the subsequent events.

The PHG is based on a human study reported by Greer *et al.* (2002). A daily oral dose of perchlorate was administered to groups of male and female volunteers at 0.007, 0.02, 0.1, or 0.5 mg/kg-day. At the end of a 14-day exposure period, reduction in 24-hr thyroidal iodide uptake occurred at all doses and was statistically significant in the three higher-dosed groups. The human data were evaluated by a benchmark dose approach, using a Hill model fitted to the data. A five percent decrease of mean radioactive iodine uptake by the thyroid was selected as the point of departure or the benchmark dose (BMD), and the lower limit of a one-sided 95 percent confidence interval on the BMD is defined as the BMDL. The estimated BMD and BMDL are 0.0068 mg/kg-day and 0.0037 mg/kg-day, respectively.

Using the identified BMDL, an overall uncertainty factor of 10, a tap water consumption rate and body weight ratio of 25.2 kg-day/L for the 95th percentile value of the pregnant woman population, and a relative source contribution of 60 percent, a health-protective concentration of 6 ppb perchlorate in drinking water can be calculated.

An uncertainty factor of 10 for interindividual variability is used because the subject population in the Greer *et al.* (2002) study is not the same as the general population. The subject population is small, consisting of 37 subjects; it did not include pregnant women,

infants, and individuals with thyroid problems. The sensitivity of the thyroid gland of fetuses and infants towards the inhibitory effect of perchlorate may be higher than that observed in adults. As iodide and perchlorate compete for the same protein receptor (NIS), the relatively high iodine intakes of the subjects in the study might have lowered the sensitivity of the thyroid towards the inhibitory effects of perchlorate. Dietary iodine intake and thyroidal iodide uptake are known to vary among individuals; they are affected by the type of food one eats (some food is rich in iodine while other foods contain goitrogens), smoking habits (tobacco smoke contains goitrogens), and exposure to environmental contaminants (such as nitrate, polychlorinated biphenyls and dioxins). The interindividual variability in the general population is likely to be greater than that in the study data.

State law requires OEHHA to account for exposure to perchlorate in media other than drinking water and there is information which indicates perchlorate can be found in some food. The Relative Source Contribution (RSC) is the proportion of the total daily exposure to perchlorate that is to be allocated to drinking water. If no other sources of the contaminant are known, then U.S. EPA recommends a value of 80 percent be allocated to drinking water. If there are other detectable but unquantifiable sources, U.S. EPA suggests a value between 20 and 50 percent of the total daily exposure be allocated to drinking water. Finally, if data exist to estimate contributions from other sources, that data can be used to calculate the source contribution.

Preliminary results have demonstrated the presence of perchlorate in some food (Kirk *et al.*, 2003; Smith and Jackson, 2003). A low level of perchlorate has also been detected in a single sample of human breast milk. While a precise value for the RSC cannot be established at this time, current scientific evidence suggests that the estimated exposure to perchlorate in water is greater than from other sources. For this reason, the RSC for this PHG is set at a level of 60 percent (instead of 20-50 percent) because OEHHA believes that the daily exposure to perchlorate would be predominantly from contaminated drinking water, not from other sources, e.g., food. Studies are underway to quantify perchlorate levels in various food types. DHS has indicated that based on a review of current research data, there is no imminent health threat from perchlorate in food that would require a change in diet.

Four sensitive subpopulations are identified in this evaluation: (i) pregnant women and their fetuses, especially those who are getting less than a sufficient amount of iodine; (ii) lactating women, especially those who are getting less than a sufficient amount of iodine, (iii) infants; and (iv) individuals with thyroid problems.

A number of human studies have shown that pregnancy itself puts stress on the thyroid (Crooks *et al.*, 1967; Glinioer *et al.*, 1990, 1992, 1995; Smyth *et al.*, 1997; Caron *et al.*, 1997; Brent, 1999; Kung *et al.*, 2000). In areas of iodine deficiency (intake level <100 µg/day), there is an increased risk of abnormally low serum triiodothyronine (T3) and thyroxine (T4) levels, thyroid enlargement, and goiter in pregnant women. The nature and severity of changes in thyroid and thyroid function are related to the severity of iodine deficiency. In three prospective studies (Romano *et al.*, 1991; Pedersen *et al.*, 1993; Glinioer *et al.*, 1995), it was shown that the pregnancy-related thyroid enlargement could be prevented by administering iodide salts to the pregnant women. These results

confirm that iodine deficiency is the main causative factor of thyroid enlargement during pregnancy.

Urinary iodine concentrations are often used as an indicator of the adequacy of iodide intake of a population. According to the National Health and Nutrition Examination Surveys (NHANES) III data, the median of the population was 14.5 µg/dL and 11.7 percent of the population was found to have low urinary iodine concentrations (<5 µg/dL) (Hollowell *et al.*, 1998). The median urinary iodine concentrations in iodine-sufficient populations should be greater than 10 µg/dL, and no more than 20 percent of the population should have urinary iodine concentration less than 5 µg/dL. If one applies this definition, there appears to be no iodine deficiency in the general population. However, the data do not preclude the possibility that some women of childbearing age are not getting the sufficient daily amount of iodide.

More significantly, developing fetuses and neonates are considered particularly sensitive to iodine deficiency because of the irreversible changes that can occur during this period of rapid structural and behavioral development. Severe iodine deficiency (<25 µg/day) during pregnancy can cause prenatal death and cretinism (WHO, 1994; as cited in Hollowell *et al.*, 1998). Cretinism is characterized by mental deficiency, deaf mutism and spastic diplegia¹. Even less than severe iodine deficiency during pregnancy has been linked to adverse neuropsychological development and a reduction of intelligence quotient (IQ) of the child (Vermiglio *et al.*, 1990; Bleichrodt and Born, 1994). Studies indicate that hypothyroidism or small decrements in maternal free T4 within what is generally considered the "normal" range (lowest 10 percent) during the first trimester are associated with impaired neuropsychological development in the child (Pop *et al.*, 1999, 2003; Haddow *et al.*, 1999; Klein *et al.*, 2001).

Adverse effects of perchlorate exposure on the development of fetal brain have also been observed in rats. In a study reported by Argus Research Laboratories (2001), ammonium perchlorate in water was administered to female rats throughout the gestation period at target doses of 0, 0.01, 0.1, 1 or 30 mg/kg-day. Thyroid weight changes were reported in the dams and pups at the higher doses. Thyroid hormones were decreased in rat fetuses and pups at various doses, some decreases were statistically significant even at the lowest dose. Several brain areas (e.g., corpus callosum) in the treated male and female pups varied significantly from controls in size. The design of the study and the way pups' brain were sectioned has been criticized. Interpretation of the study is also made difficult due to the unusual dose-response relationships of the brain morphometry data.

NIS is known to be present in the placenta and lactating mammary glands. At sufficiently high doses, perchlorate can impact the NIS in these tissues and reduce the iodine supply to fetuses and breast-fed infants. Since breast milk is the sole source of iodine for some infants, and iodine is necessary for normal thyroid function and brain development, an adequate level of iodine in breast milk is vital to the well-being of breast-fed infants. Furthermore, if there were a reduction in thyroidal iodide uptake, neonates are less capable of maintaining normal thyroid hormone production by using the iodide stored in the thyroid. Van de Hove *et al.* (1999) demonstrated that newborns have

¹ Bilateral paralysis of both sides of any part of the body.

less thyroid hormone stored in the thyroid gland and the turnover rate of the intrathyroidal pool of thyroid hormones in newborns is much higher than that in adults.

Individuals with hypothyroidism have impaired thyroid function; a significant reduction in thyroidal iodide uptake may exacerbate the illness. For this reason, they are also identified as a sensitive subpopulation.

In their latest draft perchlorate risk assessment, U.S. EPA (2002) identified a point of departure of 0.0085 mg/kg-day, based on the anti-thyroid effects of perchlorate observed across a range of experimental designs, including animal and human studies. U.S. EPA (2002) applied an overall uncertainty factor of 300 to calculate a draft oral reference dose (RfD) of 0.00003 mg/kg-day, or 0.03 µg/kg-day. Currently, there is no Federal or State maximum contaminant level (MCL) for perchlorate. The current California State drinking water action level of 4 ppb (DHS, 2004a) is being changed to 6 ppb.

INTRODUCTION

Many adverse effects observed in test animals and humans exposed to perchlorate are related to the disruption of thyroid hormone regulation. The principal hormones secreted by the thyroid are thyroxine (T4) and triiodothyronine (T3); both hormones are iodine-containing amino acids. While T4 is produced only by the thyroid gland, about 80 percent of T3 is formed in the peripheral tissues by deiodination of T4. T4 and T3 influence the growth and maturation of tissues, cell respiration and total energy expenditure, and the turnover of essentially all substrates (including carbohydrates, cholesterol, and proteins), vitamins, and hormones (including the thyroid hormones themselves).

The major substrates for thyroid hormone synthesis are iodide and tyrosine. Tyrosine is generally not the rate-limiting substrate. Iodine is a trace element and can be rate-limiting in thyroid hormone synthesis. Ingestion is the main route of iodine intake. Once ingested, iodine is reduced to iodide (I⁻) in the gastrointestinal tract and is readily absorbed into the bloodstream through either passive diffusion or active transport.

Thyroid tissue has a special ability to concentrate iodide selectively from the surroundings where the concentration is very low (10^{-8} to 10^{-7} M). The thyroid can actively transport iodide into the organ such that intrathyroidal iodide concentrations can be several hundred-fold higher than those of the external medium. Such concentrations are presumably required to promote efficient hormone synthesis because patients lacking the ability to concentrate iodide have goiters and are hypothyroid (Wolff, 1998). A protein model has been proposed of the manner in which sodium and iodide ions are actively transported into the thyroid gland. The molecule that is responsible for this transport has been named the sodium-iodide symporter (NIS). The structure and regulation of NIS have been characterized (de la Vieja *et al.*, 2000). Recently mouse symporter molecule has been cloned and expressed into normally noniodide-transporting cells; after the modification, these cells showed perchlorate-sensitive iodide accumulation capability (Perron *et al.*, 2001). These researchers found evidence to indicate that sodium-iodide symporter is also present in extra-thyroidal tissues, such as stomach and lactating mammary gland, and to a lower extent in small intestine, skin, and brain.

In humans, a majority of T4 and T3 in plasma is protein-bound. In normal plasma, the T4 binding distribution is 80 percent of T4 binding to thyroxine-binding globulin, 15 percent to transthyretin, and 5 percent to albumin and lipoproteins. For T3, the distribution is 90 percent bound to thyroxine-binding globulin and the rest to albumin and lipoproteins, with little binding to transthyretin. Very small proportions of T4 and T3 are free in plasma, 0.03 and 0.3 percent, respectively. Only the free hormone enters cells, exerts its biologic action, and determines thyroid physiologic status (Dillmann, 2000).

Control of T4 and T3 concentrations in blood is mainly regulated by a negative feedback involving three organs: the thyroid gland, which produces thyroid hormones; and the pituitary gland and hypothalamus, which respond to and help maintain optimal levels of thyroid hormones (Figure 1). The hypothalamus stimulates the pituitary through thyrotropin-releasing hormone (TRH) to produce thyroid-stimulating hormone (TSH), which then prompts the thyroid gland to produce the thyroid hormones T4 and T3. The stimulated thyroid actively transports inorganic iodide into the follicular cell and converts it to an organic form and then into thyroid hormone molecules, which can influence target organs throughout the body. The secreted T4 and T3 are metabolized in the liver and other tissues. Some thyroid hormone derivatives are excreted in the bile, and some of the iodine in them is reabsorbed. Cells in the hypothalamus and pituitary gland respond to levels of circulating thyroid hormones, i.e., when hormone levels are high, there is a signal to reduce the output of TRH and TSH. Similarly, when thyroid hormone levels are low, the pituitary is prompted to deliver more TSH to the thyroid gland to increase thyroid hormone output. This negative feedback loop helps the body to respond to varying demands for thyroid hormone and to maintain hormone homeostasis. Circulating T4, T3, and TSH can readily be monitored in the serum of experimental animals and humans by radioimmunoassay and serve as biomarkers of exposure and effect of agents that disrupt thyroid-pituitary status (U.S. EPA, 1998a, and 1998b; Hill *et al.*, 1989).

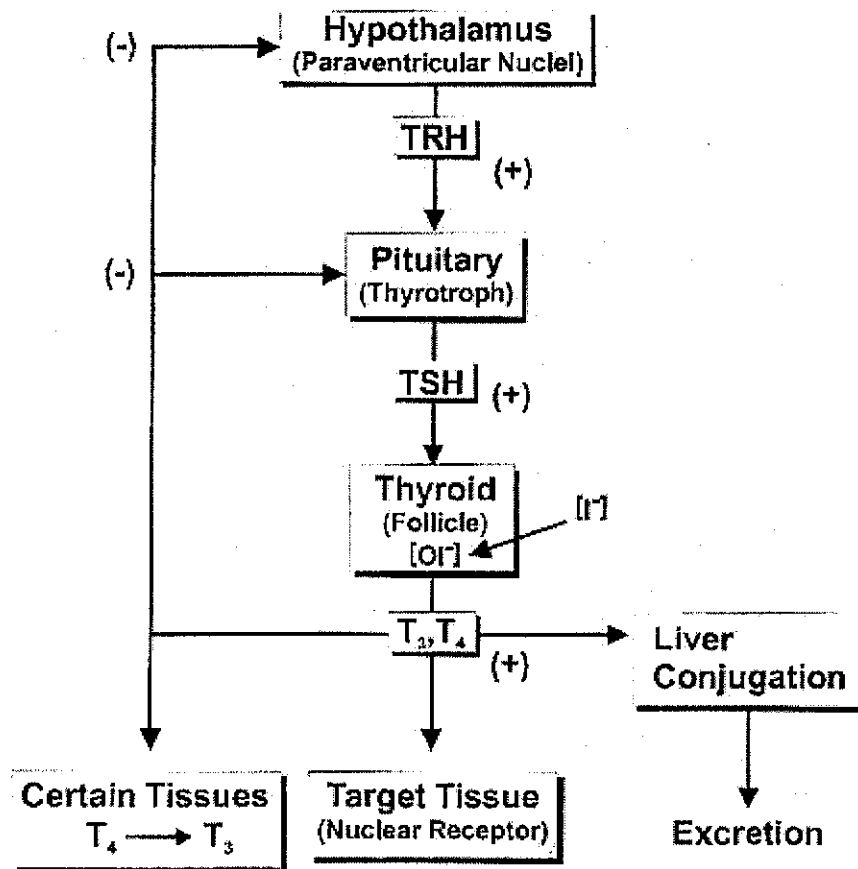


Figure 1. Hypothalamic-pituitary-thyroid axis (from U.S. EPA, 1998b)

In mammals, when demands for more thyroid hormone are small, existing thyroid follicular cells can meet the demand. With increased need, as a result of certain chemical exposures or chronic iodine deficiency, the thyroid responds by increasing the size (hypertrophy) and number (hyperplasia) of thyroid follicular cells to enhance hormone output. With continued TSH stimulation, there is actual enlargement of the thyroid gland (goiter) and, at least in rodents, neoplasia of the thyroid follicular cells could eventually occur. Since TSH-producing pituitary cells are also stimulated, they too sometimes undergo hyperplasia and neoplasia (U.S. EPA, 1998b).

Too much or too little thyroid hormone can lead to illnesses. Thyrotoxicosis occurs when tissues are exposed to excess amounts of thyroid hormones, resulting in specific metabolic changes and pathophysiologic alterations in organ function. The most frequent cause of thyrotoxicosis is Graves' disease, accounting for 60 to 90 percent of cases (Dillmann, 2000). Graves' disease is an autoimmune disorder with B-lymphocytes producing immunoglobulins, some of which bind to and activate the TSH receptor, stimulating excess thyroid growth and hormone secretion. Hypothyroidism results from decreased secretion of thyroid hormone from the thyroid gland; it can be caused by destruction of thyroid tissues or defects of thyroid hormone biosynthesis (e.g., congenital enzyme defects, congenital mutations in TSH receptor, iodine deficiency or excess). In some rare occasions, hypothyroidism can also be caused by pituitary or hypothalamic diseases.

The most severe neurologic impairment resulting from iodine deficiency is cretinism caused by iodine deficiency. Characteristics of cretinism include mental retardation, spastic dysplasia, and problems with gross and fine motor control. In some extreme forms, the affected individuals cannot walk or stand. A number of studies indicate that even less severe iodine deficiency can reduce maternal serum thyroid hormone levels and may subsequently impair the brain development of the offspring (Glorieux *et al.*, 1988; Rovet *et al.*, 1987; Tillotson *et al.*, 1994; Vermiglio *et al.*, 1990; Pop *et al.*, 1999, 2003; Haddow *et al.*, 1999; Bleichrodt and Born, 1994). The nature and severity of the adverse effects are related to the degree of iodine deficiency or the extent of maternal thyroid hormone depression.

It has been suggested that fetal damage during development is inversely related to maternal serum T4 levels in the first and second trimesters. Maternal serum free T4 is able to pass through the placenta and is converted to T3 in the fetal brain. The T3 generated *in situ* is believed to be necessary for the development of brain, specifically the cerebral cortex, the extrapyramidal system, and the cochlea (Porterfield, 2000). The availability of a minimum level of maternal free T4 is crucial for proper fetal brain development in the first and second trimesters, as the fetal thyroid is not fully mature and functional during that time period. Figure 2 shows the approximate timing of major insults to the brain resulting from hypothyroxinemia (a low level of serum T4), superimposed on major neurodevelopmental events.

It is important to note that a normal level of maternal T3 does not seem to prevent the potential damage of a low supply of T4. This may be explained by the fact that placenta is rich in Type III deiodinase which catalyzes the conversion of T4 to reverse T3 and T3 to 3,3'-diiodothyronine (T2). Both reverse T3 and T2 are not biologically active. A

majority of the T3 in fetal brain is derived from the local generation of T3 from T4 by Type II deiodinase.

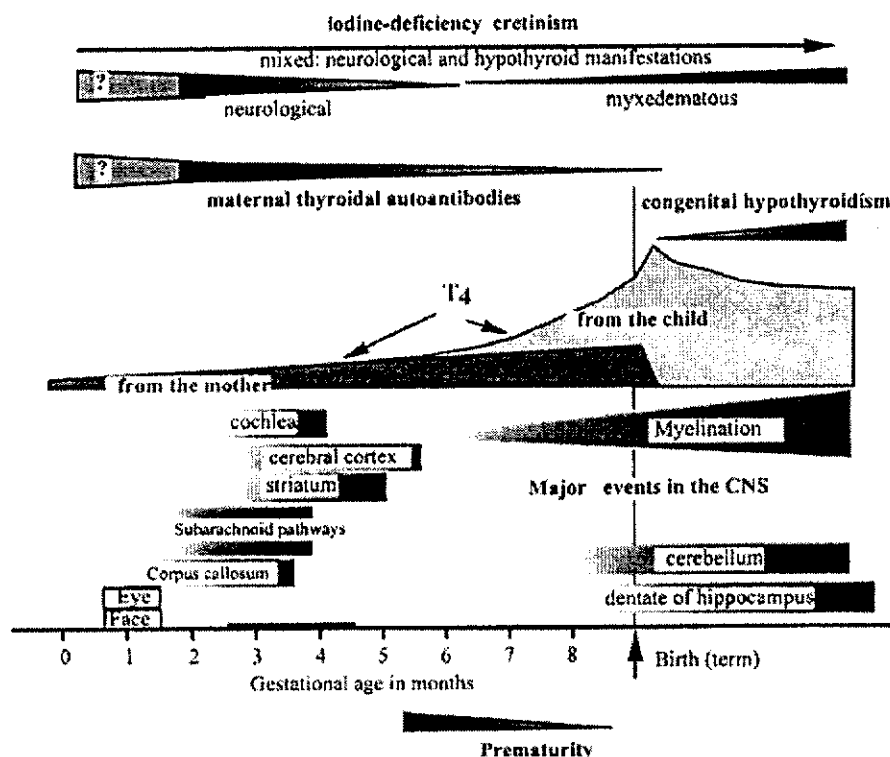


Figure 2. Approximate timing of major insults to the brain resulting from hypothyroxinemia, superimposed on major neurodevelopmental events. Conditions resulting in early maternal hypothyroxinemia, combined with later impairment of the fetal thyroid, are the most damaging, with central nervous system (CNS) damage that is irreversible at birth. The most frequent cause is maternal iodine deficiency and the presence of maternal autoimmune thyroid disorders. Unless iodine deficiency is also present, the CNS damage in congenital hypothyroidism is preventable by early postnatal treatment because the normal maternal thyroxine level has avoided damage to the brain until birth. However, normal maternal concentrations of T3 with low T4 do not protect the fetal brain because of its dependence on intracerebral regulation of local T3 availability by deiodinating pathways using T4 as a substrate. Interruption of the contribution of maternal T4 in premature infants with an immature thyroid may also underlie their increased risk of neurodevelopmental problems. The question mark indicates that we do not know whether very early CNS development, corresponding to a period when the general morphogenesis of the prosencephalon (neurulation and segmentation) is being determined, is thyroid hormone sensitive or not (from Morreale de Escobar *et al.*, 2000).

CHEMICAL PROFILE

Chemical Identity

Perchlorate (ClO_4^-) is the most oxygenated member of a series of four anions made up of chlorine and oxygen. The anion has a charge of negative one, and can form an acid or a salt in combination with H^+ or another cation such as sodium, potassium or ammonium ion. Perchlorate salts are ionic, and dissociates completely when dissolved in water. This risk assessment is for the perchlorate anion in water, regardless of the identity of the cation.

Physical and Chemical Properties

Ammonium perchlorate (NH_4ClO_4), the salt used as an oxidizer in rocket propellants, is a white, crystalline solid. As ammonium perchlorate is the major source of most of the perchlorate that has been detected in drinking water sources in California and Nevada (U.S. EPA, 1998a), it is used as the model compound to illustrate some of the physical and chemical properties of perchlorate salts (Table 1).

Table 1. Physical and Chemical Properties of Ammonium Perchlorate (from HSDB, 2000)

Property	Value or Information
Molecular Weight	117.49
Color/Physical State	white orthorhombic crystals
Melting Point	starts to decompose at 439°C
Solubility in water	200 g/L at 25°C
Solubility in organic solvents	soluble in methanol, slightly soluble in ethanol and acetone, almost insoluble in ethyl acetate, ether
Density	1.95 g/cm ³

Production and Uses

Ammonium perchlorate has been and continues to be used as an oxidizer in solid rocket propellant. Sodium perchlorate is used in slurry explosives, and potassium perchlorate is used in road flares and air bag inflation systems.

The manufacture of perchlorate salts begins with the electrolysis of brine (sodium chloride in water) to first form sodium chlorate (NaClO_3) and then sodium perchlorate

(NaClO_4). The sodium perchlorate is reacted with ammonium chloride to form ammonium perchlorate (NH_4ClO_4) and sodium chloride. The solution is cooled, and the ammonium perchlorate crystals are dried and packaged.

Ammonium perchlorate is mixed with metallic aluminum in a synthetic rubber base to make rocket fuel. This type of fuel is used in the Minuteman missile, which has been deployed in the United States since 1961. Perchlorate salts are also used as a component of air bag inflators, in nuclear reactors and electronic tubes, as additives in lubricating oils, in tanning and finishing leather, as a mordant for fabrics and dyes, and in electroplating, aluminum refining, rubber manufacture, and the production of paints and enamels (U.S. EPA, 2002).

ENVIRONMENTAL OCCURRENCE AND HUMAN EXPOSURE

Perchlorate salts have been widely used as an oxidizer in solid propellants for rockets and missiles since the mid-1940s. Because of its finite shelf life, the propellant containing perchlorate has been periodically washed out of the United States' missile and rocket inventory to be replaced with a fresh supply (U.S. EPA, 1998a). As a consequence of this use, large volumes of perchlorate have been disposed of since the 1950s. Some of this has leached into soil, and into aquifers used as drinking water sources. Perchlorate is highly mobile in aqueous systems and can persist for many decades under typical ground and surface water conditions (U.S. EPA, 1998a).

Air

Some unreacted perchlorate is occasionally released to the atmosphere during the launch of solid fuel rockets. Releases to air also occurred as a consequence of open burning and open detonation of old rocket fuel or surplus materials. No data were found on levels of perchlorate in ambient air.

Perchlorate dust can also be suspended in the air, and can be inhaled by individuals working in areas where perchlorate is manufactured (Lamm *et al.*, 1999).

Soil

Because of the finite shelf life of perchlorate used in rocket fuel, large volumes of rocket fuel containing perchlorate have been periodically washed out of the United States' missile and rocket inventory to be replaced with a fresh supply (U.S. EPA, 1998a). Releases to the environment might also have occurred because of the past open burning and open detonation of perchlorate-containing material. As a result of past disposal practices, soil and groundwater near the facilities that had been engaged in rocket fuel manufacturing and disposal are contaminated. Another way in which soil can become contaminated is by irrigation with perchlorate-contaminated water.

A report of TRC Environmental Corporation (1998) raised the concern that some chemical fertilizer may be contaminated with perchlorate. In the past, some fertilizers derived from Chilean caliche (a natural perchlorate source) were found to be

contaminated with perchlorate. Since the discovery, the producer of Chilean caliche has changed its practice and eliminated the potential perchlorate contamination. U.S. EPA (2001) tested a variety of fertilizers collected from representative sites around the nation and did not find perchlorate contamination to be a problem.

Water

Drinking water sources have become contaminated with perchlorate as a consequence of soil pollution in areas where solid rocket fuel has been used or disposed of. Perchlorate salts are soluble in water and once dissolved, perchlorate ion can persist in surface and ground waters for several decades (U.S. EPA, 1998a).

Until March of 1997, the detection limit for perchlorate in water was rather high, at 400 µg/L (ppb). In March 1997, California Department of Health Services (DHS) developed a more sensitive analytical procedure, using ion chromatography, and achieved a detection limit in the 4-5 ppb range (DHS, 2000). Shortly thereafter, the new technology was adopted by a number of commercial laboratories. EPA Method 314.0 (Federal Register, 2000) now exists for analysis of perchlorate in water to achieve a detection level as low as 0.5 ppb. The reporting level of the method is 4 ppb, however.

Because of the lack of a drinking water regulation for perchlorate, DHS established an action level of 18 µg/L (ppb) for perchlorate in 1997. In early 2002, based on a draft risk assessment on perchlorate released by U.S. EPA (2002) and the detection limit that can be generally achieved, DHS (2004a) reduced the action level to 4 µg/L (ppb). It is now being changed to 6 ppb.

Since March 1997, DHS has sampled thousands of drinking water systems and drinking water wells throughout California. As of February 2004, perchlorate detections have been reported in 348 drinking water sources (Table 2). They are primarily groundwater wells, although sources containing water from the Colorado River have also been contaminated (DHS, 2004b). DHS recommended that a source be removed from service when perchlorate concentrations were more than 10 times the action level. Perchlorate at greater than 40 µg/L (ppb) occurred in 24 sources in these counties: Los Angeles (8 sources, 47-159 ppb), Riverside (4 sources, 45-65 ppb), Sacramento (5 sources, 72-400 ppb), and San Bernardino (7 sources, 46-820 ppb).

Table 2. Reported Perchlorate Detections in California (from DHS, 2004b) ^{a, b, c, d}

County	Number of Sources	Number of Systems	Peak Level (µg/L)
Los Angeles	135	37	159
San Bernardino	82	17	820
Riverside	65	8	65
Orange	32	10	10.7
Sacramento	13	3	400
Tulare	8	4	11
Santa Clara	9	6	8.5
Ventura	2	1	20
San Diego	1	1	4.7
Sonoma	1	1	5
Total	348	88	--

^a Data are draft, and represent results from over 6,400 drinking water sources from more than 1,100 public water systems sampled since 1997.

^b This table presents sources with more than one perchlorate detection in the DHS database, where "sources" may include both raw and treated sources, distribution systems, blending reservoirs, and other sampled entities. Data do not include agricultural sources, monitoring wells, or more than one representation of the same source (i.e., a source with both a raw and treated entry, distribution system or blending reservoir is counted as a single source).

^c DHS uses a detection limit for purposes of reporting of 4 ppb. Detections at or above the limit are required to be reported. Some laboratories may report data at lower concentrations.

^d As of February 4, 2004

Urbansky *et al.* (2000) analyzed samples of eight domestic brands and eight imported brands of bottled water and did not find perchlorate (with a detection limit of 5 ppb) in any of the samples.

Food

It has been known for a long time that perchlorate can be used as a growth promoter in leguminous plants (Verteletskaia *et al.*, 1974; as cited in Von Burg, 1995), livestock (sheep and cattle) and poultry (Yakimenko *et al.*, 1981; as cited in Von Burg, 1995). The cited USSR research indicated that weight gains in test livestock of 3-31 percent were obtained by addition of ammonium perchlorate to the feed. Feed expenditure was also reduced 7-18 percent. The optimum dose was estimated to be 2-5 mg/kg (Grayson, 1978). Weight gains in livestock may be secondary to hypothyroidism, decreasing their metabolic rates.

There are preliminary results showing that some vegetables (e.g., lettuce) may bioconcentrate perchlorate. In a greenhouse study, U.S. EPA researchers watered lettuce plants with one of five different concentrations of perchlorate (0.1, 0.5, 1.0, 5.0, and 10.0 $\mu\text{g}/\text{mL}$) for a period of 90 days following planting. They found perchlorate levels rose steadily over the first 50-60 days, and then generally leveled off. The amount of perchlorate detected in the leaves correlates with dose. For example, at about 50 days into the study, the lettuce irrigated with 10.0 $\mu\text{g}/\text{mL}$ (ppm) perchlorate exhibited a perchlorate content of about 300 $\mu\text{g}/\text{g}$ on a wet mass basis (U.S. EPA, 2002). As recently reported in an unpublished study, perchlorate has been detected in a number of farm produce samples, such as lettuce, cucumbers, strawberries, and blackberries (Smith and Jackson, 2003). In another unpublished study, USDA detected perchlorate in 6 out of 10 lettuce samples tested. In the detected samples, perchlorate levels ranged from 46 to 81 ng/g fresh weight (personal communication with DHS, 2003).

There are concerns that breast milk may represent an exposure pathway for infants. In a study reported by Yu (2000), groups of female rats were treated with perchlorate in drinking water at 0, 0.01, 0.1, 1, and 10 mg/kg-day throughout gestation and lactation. On postnatal day 10, the rats were milked. Yu found the levels of perchlorate in milk were about twice as high as the corresponding levels in maternal serum across all doses, suggesting that perchlorate is actively sequestered into milk. Clewell *et al.* (2003) reported that perchlorate was indeed transferred to the pup through suckling as perchlorate was detected in milk, as well as in the neonate serum, gastrointestinal contents, and skin.

Human mammary gland during lactation has been shown to express the NIS and may have the capability to actively secrete perchlorate into the breast milk (Vayre *et al.*, 1999; Tazebay *et al.*, 2000). There is very little information available regarding the secretion of perchlorate into breast milk by nursing mothers, but this type of information is available for pertechnetate (TcO_4^-). Iodide and pertechnetate have similar ionic characteristics and are actively taken up by the NIS (Zuckier *et al.*, 2001). Actually, it has been suggested that pertechnetate has a greater ability than perchlorate in inhibiting iodide transport into thyroid tissues *in vitro* (Wolff, 1964; as cited in Wolff, 1998). In three clinical case studies (Wayne *et al.*, 1986; Mountford and Coakley, 1987; Ahlgren *et al.*, 1985), $^{99\text{m}}\text{TcO}_4^-$ was injected into nursing mothers (one patient in each study) to obtain scintigraphs of the thyroid. Following the injection, breast feeding was stopped for at least 24 hr and the breast milk was collected for analysis. As the biological half-life of $^{99\text{m}}\text{TcO}_4^-$ is estimated to be approximately 3-5 hr, most of the injected dose would be excreted within 24 hr. They found the total amount of radioactivity in the milk ranged from 3 to 11 percent of the injected $^{99\text{m}}\text{TcO}_4^-$ dose.

Kirk *et al.* (2003) analyzed seven milk samples purchased from a supermarket in Texas and found perchlorate levels ranging from 1.7 to 6.4 $\mu\text{g}/\text{L}$. While a majority of the milk was bottled locally at the same plant, two were bottled in separate plants located outside the State of Texas. They also analyzed the local tap water and reported a mean perchlorate level of 2.5 ± 1.1 $\mu\text{g}/\text{L}$ in the samples that were above a detection limit of 0.5 $\mu\text{g}/\text{L}$. Kirk *et al.* (2003) analyzed a breast milk sample from a local human volunteer and found 3-4.5 $\mu\text{g}/\text{L}$ of perchlorate using two different analytical methods. While these data

strongly suggest consumptions of cow's milk and human breast milk are viable exposure pathways, the data also indicate perchlorate may not be highly concentrated in milk.

Another concern is that iodide in breast milk is necessary for thyroid hormone synthesis by the newborn. Perchlorate inhibits the NIS in the lactating mammary gland and can interfere with the secretion of iodide into breast milk. This reduction in iodide transfer has been seen in cows and goats (Howard *et al.*, 1996; Lengemann, 1973; Mountford *et al.*, 1987). However, if it is assumed that the properties (i.e., relative affinities for iodide and perchlorate) of NIS in the lactating mammary gland are similar to those of NIS in the thyroid, a dosage of perchlorate that is not expected to inhibit thyroidal iodide uptake would probably not inhibit the excretion of iodide into the breast milk.

METABOLISM AND PHARMACOKINETICS

Absorption

Test data from human studies indicated that perchlorate is readily absorbed from the gastrointestinal tract and excreted primarily via the urine. Eichen (1929; as cited in Stanbury and Wyngaarden, 1952) administered orally 1-2 g perchlorate to patients and recovered 70 percent of the dose in the urine in 12 hr and 85-90 percent in 24 hr. In a similar experiment, two human subjects each drank a solution of 794 mg of sodium perchlorate dissolved in 100 mL of water (Durand, 1938). Fifty percent of the administered dose was recovered in the urine by five hours and 95 percent in 48 hours. These human data suggest absorption of perchlorate through the oral route is virtually complete.

Besides the thyroid, the NIS appears to be expressed and active in mammary gland, salivary glands, gastric mucosa, and placenta (Vayre *et al.*, 1999; Tazebay *et al.*, 2000; de la Vieja *et al.*, 2000; Mitchell *et al.*, 2001). These transport systems exhibit functional similarities with their thyroid counterpart and may play a role in the absorption of iodide into the body.

Because perchlorate is completely ionized in aqueous systems, its permeability through intact skin is expected to be limited (U.S. EPA, 1998a). Inhalation exposure during showers is considered possible but not likely to be an important route of exposure. This is because the droplets produced in showers are generally too large to be inhaled. Exposure to vapors of the chemical via the inhalation route is expected to be negligible because of the low vapor pressure of perchlorate salts at room temperature. However, inhalation of airborne perchlorate particles could be an important exposure route in occupational settings. Lamm *et al.* (1999) studied a group of workers in a perchlorate production plant and reported that there was a correlation between airborne perchlorate dust concentration and the amount of perchlorate excreted in urine.

Distribution

Anbar *et al.* (1959) injected white rats and rabbits intraperitoneally with radiolabeled potassium perchlorate (approximately 3-14 mg per animal) and measured the specific

activity per gram of tissue in various organs from 30 minutes to 12 hours post administration. The ratio of the specific activity of perchlorate in thyroids versus the specific activity in blood reached a limiting value of 4.3 ± 0.3 in both rats and rabbits, at about 6 hours after the injection. These data demonstrate the fact that thyroid concentrates perchlorate ions in rat and rabbit. There were also indications of retention of perchlorate in the salivary gland and in the testes, when the perchlorate level in blood fell to a very low level.

Chow *et al.* (1969) compared the uptake of radiolabeled perchlorate with various amounts of stable perchlorate in male Sprague-Dawley rats. The male rats were injected with 0.1, 0.2, or 5.0 meq/kg perchlorate (14, 28, or 690 mg/kg, respectively) 2 hr prior to sacrifice. It was found that at the low and middle doses, radiolabeled perchlorate concentrations in the thyroid were higher than those in the blood. At the high dose, perchlorate concentrations in the thyroid and blood were about the same. In a similar study, rats were exposed to 0.69, 1.4, 2.8, 6.9, or 14 mg/kg. The apparent accumulation of perchlorate in the thyroid, as reflected in the thyroid/blood ratio (ranged from 31.1 to 2.5), was found to be inversely related to the perchlorate dose (U.S. EPA, 2002).

Chow and Woodbury (1970) also studied perchlorate accumulation by the thyroid at low doses. They administered perchlorate by intraperitoneal injection at 0.69, 14, or 280 mg/kg to groups of male Sprague-Dawley rats. The treated rats were sacrificed at 0.033, 0.067, 0.13, 0.2, 0.5, 1.0, 2.0, and 4.0 hr after dosing. The levels of perchlorate measured in the thyroid and plasma indicate that perchlorate was best accumulated in the thyroid at the low dose. At higher doses (at or above 14 mg/kg), the level of perchlorate in the thyroid was lower than in the plasma.

It has been shown that perchlorate inhibits iodide transport into the thyroid. This is because in addition to iodide, thyroid tissues can also select several related monovalent anions. Measurement of the ability to be concentrated by thyroid tissues, or to inhibit iodide transport, has resulted in the following potency series for monovalent anion-based inhibition of iodide transport in thyroid slices: $\text{TcO}_4^- \geq \text{ClO}_4^- > \text{ReO}_4^- > \text{SCN}^- > \text{BF}_4^- > \text{I}^- > \text{NO}_3^- > \text{Br}^- > \text{Cl}^-$ (Wolff, 1964; as cited in Wolff, 1998). It is not clear whether this anion sequence, measured at very high concentrations, has any necessary mechanistic relation to what occurs in the thyroid, at low concentrations. Anbar *et al.* (1959) showed that the inhibition of iodide transport by perchlorate is a truly competitive process. They injected intraperitoneally ^{36}Cl -labeled perchlorate and iodide ions in various concentrations to groups of rats and found that either iodide or perchlorate could inhibit the accumulation of the other anion by the thyroid (Table 3).

Recently, the apparent accumulation of perchlorate by the thyroid of rodents has been disputed. Citing *in vitro* electrophysiological data, de la Vieja (2000) suggested perchlorate acts as a blocker of NIS, but it is not translocated via NIS into the cell. De la Vieja (2000) theorized that because ^{36}Cl chlorate (ClO_3^-) is a byproduct of the reaction employed to chemically synthesize ^{36}Cl perchlorate for the uptake study, it is possible that ^{36}Cl chlorate, rather than perchlorate, accounts for the measured radioactivity, given that chlorate is readily translocated via NIS into the cell.

Table 3. The Ratio between Concentrations of Iodide and Perchlorate Ions in the Thyroid (from Anbar *et al.*, 1959)

Iodide dose (mmol)	Perchlorate dose (mmol)	Ratio* I/ClO_4^-		
		60 min	120 min	360 min
0.14	0.028	4.7	6.9	3.3
0.14	0.14	2.1	2.7	2.2
0.028	0.14	0.53	0.58	0.67

*ratio = concentration of iodide in thyroid / concentration of perchlorate in thyroid

Goldman and Stanbury (1973) administered ^{36}Cl -labeled potassium perchlorate to male Sprague-Dawley rats by intraperitoneal injection (approximately 40 μg stable perchlorate per injection). The rats were maintained on a low iodide diet for 4-5 weeks prior to the time of perchlorate administration. The level of perchlorate in the thyroid peaked at four hours after administration, then declined to approximately five percent of its peak value at 96 hours. The decay followed an exponential function with a half-life of 20 hr. When the levels of radioactivity in the serum and the urine are plotted against time, they also follow an exponential function with a half-life of approximately 20 hr. Goldman and Stanbury (1973) also showed that most of the administered perchlorate was excreted in the urine. The retention of the radiolabel in selected tissues 96 hr after the administration of perchlorate is shown in Table 4.

There are also data indicating that perchlorate can pass through the placenta and affect the fetal thyroid. Thyroid enlargement and reduction of thyroidal iodide uptake have been detected in fetuses of laboratory animals exposed to perchlorate (see the discussion in the Developmental and Reproductive Toxicity section.)

Table 4. Percent Dose of $^{36}Cl/g$ Tissue at 96 Hour after Intraperitoneal Injection of ^{36}Cl -perchlorate (from Goldman and Stanbury, 1973)

Organ	Percent dose/g tissue
Thyroid	0.142 \pm 0.1 *
Kidney	0.125 \pm 0.09
Spleen	0.098 \pm 0.03
Liver	0.048 \pm 0.04
Brain	Background

* mean \pm standard deviation; each value represents five animals.

Selivanova and Arefaeva (1986) administered a single oral dose of perchlorate to rats and observed a two-phase biological decay curve. The first biological half-life ranged from 1-2 hr and accounted for a calculated 96 percent of the dose. The second-phase half-life,

which accounted for only 4 percent of the administered dose, ranged from 72 to 80 hr. Yu *et al.* (2000) injected perchlorate intravenously at 0.01, 0.1, 1, or 3 mg/kg to male Sprague-Dawley rats and monitored the serum concentration of perchlorate over time. In this study, the estimated early- and terminal-phase half-lives of perchlorate in rat were 2-3 hr and 12-26 hr, respectively.

The differences in biological half-lives of perchlorate in rats in the studies above (Goldman and Stanbury, 1973; Selivanova and Arefaeva, 1986; Yu *et al.*, 2000) are partly due to the three different routes of administration. The difference may also be explained by the fact that the rats in the study reported by Goldman and Stanbury (1973) had been maintained on a low iodide diet for 4-5 weeks before the administration of perchlorate.

Metabolism

There are data to suggest that perchlorate is not metabolized in humans (Anbar *et al.*, 1959). Four patients were orally administered 200 mg of radiolabeled perchlorate (5 μ Ci), double labeled with ^{36}Cl and ^{18}O . The perchlorate was excreted unchanged in the urine with the two labels (^{36}Cl and ^{18}O) remaining associated in the same molecule. The results also showed that there was no reduction of perchlorate *in vivo*, as there was very little radioactivity associated with Cl^- and ClO_3^- ions in urine.

Excretion

As described above, 95 percent of a dose of sodium perchlorate administered orally to human subjects was eliminated in the urine by 48 hours after administration (Durand, 1938). Lamm *et al.* (1999) monitored urinary perchlorate levels of two workers during three days with measurable occupational perchlorate exposure and during the subsequent three days without known perchlorate exposure. The perchlorate body burden, as indicated by urinary perchlorate concentration, increased over the three days of work exposure, with a decrease between the 12-hour work shifts. The elimination of perchlorate after the last exposure period appeared to follow a first-order kinetics pattern. The average perchlorate elimination half-lives measured for the two workers were 7.9 hours and 8.2 hours.

Greer *et al.* (2002) administered oral doses of perchlorate in water to human volunteers and estimated half-life values ranged from 6.0 to 9.3 hr, with an average value of 8.1 hr. These estimates are consistent with those provided by Lamm *et al.* (1999).

Selivanova and Arefaeva (1986) administered a single oral dose of perchlorate to rats, rabbits, and calves at 2, 20, 200, and 600 mg/kg in a single oral dose. They reported that in all cases, little or no perchlorate could be detected in the blood after 72 hours. A majority of the administered perchlorate was excreted in the urine; the feces excreted \leq 8.5 percent. Yu *et al.* (2000) injected perchlorate intravenously to rats at doses of 0.01, 0.1, 1, or 3 mg/kg and reported that between 72 percent and 97 percent of the administered dose was excreted in the urine over a 24-hr period.

Physiological/Nutritional Role

Perchlorate has no known nutritional role. In 1952, investigators observed that perchlorate displaces iodide from the rat thyroid (Wyngaarden *et al.*, 1952). Since then perchlorate has been widely used in studies on the thyroid to block entry of iodide into the thyroid, or to cause discharge of noncovalently bound iodide previously accumulated in the thyroid (Wolff, 1998).

It was reported that when ammonium perchlorate was added to the feed of farm animals, weight gain was increased by 3-31 percent. The optimum dose ranged from 2 to 5 mg/kg (Grayson, 1978; Yakimenko *et al.*, 1981; as cited in Burg, 1995). This is most likely a non-nutritive effect associated with inhibition of thyroid hormone production (hypothyroidism).

TOXICOLOGY

As perchlorate competitively blocks iodide from entering the thyroid gland, many of the adverse effects of perchlorate exposure in the low dose range are expected to be similar to those of iodine deficiency. For this reason, an overview of some of the adverse health effects of iodine deficiency is provided in this section.

In areas of inadequate iodine intake, thyroid hormone synthesis and secretion both decline. The pituitary gland responds by secreting more TSH, which in turn causes thyroid hypertrophy and iodine deficiency goiter. Children who are born in areas of severe iodine deficiency may suffer from cretinism. The main cause of the disease is iodine deficiency, but it is aggravated by dietary goitrogens, selenium deficiency, and autoimmune hypothyroidism. The manifestations of endemic iodine deficiency range from goiter or mild mental retardation in euthyroid subjects to severe mental deficiency and neurologic defects in those with variable degrees of hypothyroidism. Two subtypes of endemic cretinism have been described, neurologic cretinism and myxedematous cretinism². Neurologic cretinism is more common. It is characterized by delayed growth of long bones, neurological complications such as deaf mutism, mental retardation, and spasticity. Goiter is sometimes associated with the illness, but not myxedema. Myxedematous cretinism is less common. It is characterized by delayed growth of long bones, myxedema, and sometimes goiter but with fewer neurologic problems than are seen in neurologic cretinism. Neurologic damage in the absence of neonatal hypothyroidism has been postulated to be due to maternal hypothyroxinemia early in gestation (Burrow *et al.*, 1994) (also see Figure 2).

Hypothyroxinemia or hypothyroidism during pregnancy has been linked to adverse neuropsychological development and a reduction of IQ of the child (Glorieux *et al.*, 1988; Rovet *et al.*, 1987; Tillotson *et al.*, 1994; Vermiglio *et al.*, 1990; Haddow *et al.*, 1999; Pop *et al.*, 1999, 2003; Klein *et al.*, 2001).

² Hypothyroidism leads to a slowing of metabolic processes and in its most severe form to the accumulation of mucopolysaccharides in the skin, causing a non-pitting edema termed myxedema.

The presence of maternal thyroid peroxidase antibodies (TPO-ab) during pregnancy with no abnormalities in free T4 or TSH has been found to be associated with a significant risk for reduction in scores on the McCarthy Scales of Children's Abilities. The General Cognitive Scale, which reflects IQ scores, showed a decrement of over 10 points in those with antibodies present (Pop *et al.*, 1995). At least 10 percent of women of childbearing age have these antibodies present (Weetman, 1994). In a further study, Pop *et al.* (1999) demonstrated that TPO-abs were an effective marker for free T4 in the lowest 10 percent. In that study it was found that 25 percent of women with elevated TPO-ab at 12 weeks gestation had free T4 in the lowest 10th percentile (compared with 8 percent of those negative) and of those that were antibody positive at 32 weeks, 46 percent were in the lowest 10th percentile at 12 weeks. A positive correlation was again found between being in the lowest 10th percentile of free T4 at 12 weeks gestation and risk of impaired psychomotor development at ten months of age (RR 5.8, 95 percent CI: 1.3-12.6). Pop *et al.* (2003) reported that in a prospective 3-year follow-up study of pregnant women and their children, maternal hypothyroxinemia during early gestation increased the risk of a delay in infant neurodevelopment.

These studies indicate that even small decrements in maternal free T4 within the "normal" range during the first trimester may result in impaired neuropsychological development of the child. An exact level of free T4 at which this effect would not be seen has not been determined. Women with TPO-ab during pregnancy constitute a distinct subpopulation at dramatically increased risk for free T4 < 10th percentile and subsequent impairment of their children.

It has been shown that pregnancy itself puts stress on the thyroid (Glinoe, 2001). In areas of iodine deficiency, there is an increased risk of abnormally low serum T4 and T3 levels, thyroid enlargement as well as goiter in pregnant women. The nature and severity of changes in thyroid functions are related to the severity of iodine deficiency. For this reason, pregnant women with marginal or frank iodide deficiency and their fetuses are identified as sensitive subpopulations for perchlorate exposure.

Toxicological Effects in Animals

In December 31, 1998, U.S. EPA released a draft document titled "Perchlorate Environmental Contamination: Toxicological Review and Risk Characterization Based on Emerging Information." Then in January 18, 2002, U.S. EPA updated the toxicological information and risk assessment on perchlorate and released an external review draft titled "Perchlorate Environmental Contamination: Toxicological Review and Risk Characterization." Some of the evaluations of animal toxicity studies and physiologically based pharmacokinetic modeling results described in this document are derived from the 2002 U.S. EPA document.

Acute Toxicity

In acute toxicity testing, animals generally died within the first few days after oral administration of high doses of ammonium perchlorate (750 to 4,200 mg/kg). Autopsy findings included necrosis and hemorrhaging of the mucous membranes of the stomach.

Intestinal damage, pulmonary edema, and vascular dilation and congestion of the spleen, brain and sinuses were also noted (Von Burg, 1995).

Table 5, showing acute LD₅₀ values for perchlorate salts in several species, is modified from Von Burg (1995), compiled from Schilt (1979), U.S. EPA (1971), Shigan (1963), and Joesten and Hill (1966). The lethal dose for the various perchlorate salts when administered to mice by intraperitoneal injection varied over a 50-fold range (29 mg/kg to 1,500 mg/kg), so the cation appears to play a role in the toxicity of these salts.

Mannisto *et al.* (1979) administered potassium perchlorate to male Sprague-Dawley rats in drinking water for four days at concentrations of 0, 10, 50, 100 or 500 mg/L. At the end of the exposure period, they measured blood levels of TSH and thyroid hormones (T3 and T4). Significant changes (increased TSH and decreased T3 and T4) were observed in the 100 and 500 mg/L (15.3 and 76.3 mg/kg-day) exposure groups. In the 50 mg/L (7.6 mg/kg-day) exposure group, there was a significant decrease in concentration of T3 and T4; the TSH level was increased, but the increase was not significant.

Table 5. Acute LD₅₀ Values for Perchlorate Salts (Modified from Von Burg, 1995; Schilt, 1979; U.S. EPA, 1971; Shigan, 1963; Joesten and Hill, 1966)

Species	Route of administration	Cation	Dose (mg/kg)
Rat	oral	NH ₄ ⁺	3,500 to 4,200
Mouse	oral	NH ₄ ⁺	1,900 to 2,000
Rabbit	oral	NH ₄ ⁺	750 to 1,900
Guinea Pig	oral	NH ₄ ⁺	3,310
Mouse	i.p.	Li ⁺	1,160
Mouse	i.p.	Mg ⁺⁺	1,500
Mouse	i.p.	Na ⁺	1,150
Mouse	i.p.	Mn ⁺⁺	410
Mouse	i.p.	Fe ⁺⁺⁺	370
Mouse	i.p.	Co ⁺⁺	160
Mouse	i.p.	Ni ⁺⁺	100
Mouse	i.p.	Cu ⁺⁺	29
Mouse	i.p.	Zn ⁺⁺	76

Subchronic Toxicity

In many animal studies, perchlorate has been shown to perturb thyroid hormone regulation, induce hypertrophy and hyperplasia in thyroid follicular cells, and cause an increase in thyroid weight.

Shigan (1963) administered ammonium perchlorate to "white rats" at 650 mg/kg-day for one month and did not observe noticeable cumulative properties. They also exposed "white rats" to ammonium perchlorate for three months at 190 mg/kg-day and found the treatment affected the regulation of the involuntary nervous system, caused changes in the protein fractions of the blood serum, and disrupted the liver's ability to produce glycogen for carbohydrate storage.

In a follow-up study, Shigan (1963) treated rabbits and "white rats" with 0, 0.25, 2.0, and 40 mg/kg-day of potassium perchlorate for 9 months. Many study details, such as sex, number of animals in each dose group, and dosing medium were not reported. In the two highest dose groups, the authors found a significant increase in the amount of iodide excreted from the thyroid. It is not clear if the reported effect was seen in one or both species tested (U.S. EPA, 2002).

Caldwell *et al.* (1995) administered ammonium perchlorate to groups of Sprague-Dawley rats (six males and six females per group) in drinking water for 14 days at concentrations of 0, 1.25, 5.0, 12.5, 25, 50, 125 or 250 mg/L. The corresponding doses (male/female) in mg/kg-day are 0, 0.11/0.12, 0.44/0.47, 1.11/1.23, 2.26/3.06, 4.32/4.91, 11.44/11.47, and 22.16/24.86 mg/kg-day, respectively. At the end of the exposure period, thyroids were weighed, thyroid histopathology and morphometry examinations were performed, and thyroid hormone levels were measured with a radioimmune assay technique. The researchers reported that perchlorate exposure decreased circulating T3 and T4 and increased serum TSH. There is also evidence that rT3 (formed mostly in extrathyroidal tissues) and thyroglobulin levels were also increased. They also found perchlorate exposure was associated with decreases in thyroid gland follicular lumen size and increases in relative thyroid weights. At the lowest dose, 0.1 mg/kg-day, statistically significant changes in serum T4 were observed in both sexes; this level can be identified as a LOAEL.

Springborn Laboratories (1998) administered ammonium perchlorate via drinking water to male and female Sprague-Dawley rats (10 rats/sex/dose) at doses of 0, 0.01, 0.05, 0.2, 1.0, and 10 mg/kg-day for 14 and 90 days. An additional 10 rats/sex/dose were sacrificed after a 30-day recovery period following cessation of the 90-day exposure at doses of 0, 0.05, 1.0, and 10 mg/kg-day, to evaluate reversibility of any observed lesions. No statistically significant toxicological findings were observed among the groups with respect to clinical observations, body weights, food or water consumption, ophthalmology, hematology, or clinical chemistry. The researchers reported perchlorate exposure was associated with increased thyroid weights, thyroid follicular cell hypertrophy, and thyroid colloid depletion. Mean thyroid weights of male rats in the highest dose group were significantly increased after 14 and 90 days of exposure, while mean thyroid weights of female rats in the highest dose group were significantly increased after 90 days of exposure. No thyroid pathology was observed in other dose groups. After a 90-day exposure period followed by a 30-day recovery period, there was no increase in thyroid weight in either male or female rats in the 0.05, 1.0, and 10 mg/kg-day dose groups (Siglin *et al.*, 2000).

After 14 days of exposure, mean serum TSH levels were significantly increased in males at 0.2 mg/kg-day and higher, and in females at 0.05 mg/kg-day and higher, compared with the controls. Mean serum T4 levels were significantly decreased in both sexes at 10

mg/kg-day. Mean serum T3 levels were significantly decreased in males at levels of 0.01 mg/kg-day and higher. No statistically significant differences in T3 levels were observed in the female treatment groups.

After 90 days of treatment, mean TSH levels were significantly increased in males at 0.2 mg/kg-day and higher, and in females at 10 mg/kg-day only. Mean T3 and T4 levels were significantly decreased in both sexes at 0.01 mg/kg-day and higher. Based on this data set, a LOAEL of 0.01 mg/kg-day can be identified.

Following a 90-day exposure period and a 30-day recovery period, TSH levels were significantly increased in all three female recovery groups (0.05, 1.0, and 10 mg/kg-day), whereas no significant differences in TSH levels were observed in the male recovery groups. Mean T4 levels were significantly lowered in all three male recovery groups (0.05, 1.0, and 10 mg/kg-day), whereas no significant differences in T4 levels were observed in the female groups. Mean T3 level was significantly lower in females at 10 mg/kg-day. No statistically significant differences in T3 levels were observed in the male recovery groups (Siglin *et al.*, 2000). The authors did not speculate on the differences between the male and female data.

Siglin *et al.* (2000) noted there was a change in the mean TSH levels in the male and female control groups between 14 days and 120 days. They also found changes in mean T3 levels of control females over the course of the study. It was not clear if the observed variability in mean control hormone levels was reflective of normal age-related variations or due to other factors such as the relatively small sample sizes.

Genetic Toxicity

Ammonium perchlorate was tested in a battery of genotoxicity tests, and found to be negative in all tests (U.S. EPA, 2002). Ammonium perchlorate was negative in the reverse mutation assay in *Salmonella typhimurium* (TA98, TA100, TA1535, TA1537) with and without S9 activation (ManTech Environmental Technology, 1998). The Ames tests were later repeated by the National Institute of Environmental Health Sciences. Strains TA102 and TA104 were added to cover the possibility that ammonium perchlorate causes mutation by producing active oxygen species or other DNA damaging radicals. The repeat tests also used the pre-incubation test methodology as it provides better contact between the test material and the target organism. Ammonium perchlorate was negative in the reverse mutation assay in *S. typhimurium* (TA98, TA100, TA1535, TA97, TA102, and TA104) with and without S9 activation, thus confirming the original Ames test results (Zeiger, 1998).

Ammonium perchlorate was negative in the L5178Y/TK^{+/+} mouse lymphoma assay without S9 activation. Results of the mouse lymphoma assay with S9 activation were equivocal because of low frequency of mutations in the positive controls (ManTech Environmental Technology, 1998). The test was later repeated. In this assay, concentrations of ammonium perchlorate in the treatment medium of 50 to 5,000 µg/ml were negative in the L5178Y/TK^{+/+} mouse lymphoma mutagenesis assay in the absence and presence of Arochlor-induced rat liver S9 (BioReliance, 1999). The results of the repeat study provided support for the negative results reported in the first study.

Ammonium perchlorate was tested negative in *in vivo* micronuclei assays in mice and rats. In the mouse micronucleus assay, five male and five female Swiss CD-1 mice were dosed by gavage at 0, 62.5, 125, 250, 500, or 1,000 mg/kg for three consecutive days. No increases in the frequency of micronuclei were found for any dose group (ManTech Environmental Technology, 1998). There is some uncertainty whether the maximum tolerated dose was reached in the study. Typically, the assay is performed at 85 percent of the maximum tolerated dose, and the 1,000 mg/kg-day dose represents approximately 50 percent of the LD₅₀. Furthermore, there was no indication of toxicity to the bone marrow cells because the polychromatic erythrocyte/normochromatic erythrocyte ratio was not different from the controls. The test was later repeated by the National Institute of Environmental Health Sciences. Male B6C3F₁ mice were injected interperitoneally with 0, 125, 250, 500, 1,000, 1,500, and 2,000 mg/kg ammonium perchlorate in buffered saline. Five mice per group were treated daily for three consecutive days, and were sacrificed 24 hrs after the last injection. All animals in the 1,500 and 2,000 mg/kg groups died after the first injection and 4/5 animals in the 1,000 mg/kg group died after the second injection. All animals in the 125, 250, and 500 mg/kg groups survived the treatment. No increases in micronuclei were seen at any of the test doses, and the trend test was not positive (Zeiger, 1999). The negative results of the repeat study support the results of the first study.

The 90-day subchronic bioassay using Sprague-Dawley rats also evaluated micronuclei induction (Siglin *et al.*, 2000). Ten rats per sex were treated with ammonium perchlorate in drinking water for 90 days at 10 mg/kg-day. The results indicate that ammonium perchlorate under the test condition was not mutagenic to the bone marrow cells of male and female Sprague-Dawley rats. The chemical was not toxic to the bone marrow cells at the dose tested, as it did not cause any reduction of the polychromatic erythrocyte/normochromatic erythrocyte ratio in male or female rats.

Based on the above *in vitro* and *in vivo* genotoxicity test results, ammonium perchlorate does not appear to be mutagenic or clastogenic. Therefore, genotoxicity is not considered a potential mode of carcinogenic action for perchlorate.

Developmental and Reproductive Toxicity

A number of toxicity studies have shown that perchlorate exposure causes a variety of adverse health effects in the offspring of the test animals.

Developmental toxicity

Postel (1957) gave a one percent solution of potassium perchlorate to eleven guinea pigs in the second or third week of pregnancy. The control group consisted of three pigs receiving a diet containing 0.48 µg/g and 50 mg of ascorbic acid orally twice a week. The treated pigs were divided into four groups which received 0, 8, 16, or 32 µg, respectively, of triiodothyronine (T3) by subcutaneous injection each day. Control pigs and those receiving potassium perchlorate alone were given daily injections of saline solution. The overall mean dose of potassium perchlorate was 1520 mg/kg-day in the T3-injected pigs and 740 mg/kg-day in the group receiving only potassium perchlorate. The mean exposure duration was 37 days (range 21-48 days). One hour before the removal of the fetuses, radioactive sodium iodide was injected subcutaneously to the

mother. Radioactivity of the blood and thyroids of the mothers and fetuses were analyzed.

Postel reported that massive enlargement of the fetal thyroid was observed in all the perchlorate treated groups, with or without T3 injection. The overall mean weight was 491 mg/100 g body weight, compared with a mean control fetal thyroid weight of 32 mg/100 g. Perchlorate treatment did not cause enlargement of the maternal thyroid in all the treated groups, with or without T3 injection.

Postel (1957) also reported that the thyroid/serum radioiodide concentration ratio was approximately 175 in fetus and 50 in mothers in the control group. It was suggested that the data support the concept that the normal fetal thyroid is in a relatively hyperplastic state. The thyroid/serum radioiodide concentration ratios of both the perchlorate-treated mothers and their fetuses were significantly depressed, compared to the controls.

Lampe *et al.* (1967) gave perchlorate-treated food to 12 rabbits from the beginning of pregnancy through gestation day 21 or 28. The rabbits were dosed at 100 mg/kg-day. No concurrent controls were used. Ingestion of perchlorate was found to cause an increase in maternal and fetal thyroid weights. On the 21st day of treatment, the maternal thyroid weights in treated animals were nearly three times higher than control thyroids; fetal thyroids were nearly four times the control weights. Continued intake of perchlorate further enhanced the increase in thyroid weight, particularly in the fetus. On the 28th day of treatment, the maternal thyroid weights in treated animals were nearly four times higher than control thyroids; fetal thyroids were nearly nine times the control weights. The researchers noted that the placenta is permeable to perchlorate, which caused intensive hyperplasia as a sign of hyperfunction, and a loss or complete disappearance of colloid in the maternal and fetal thyroids. They also noted the increased sensitivity of fetal thyroid to the treatment, compared with the maternal thyroid.

Sztanyik and Turai (1988) investigated the safety of using potassium perchlorate or potassium iodide as blocking agents to prevent the uptake of radioiodine by fetuses. They injected these compounds to pregnant albino rats (body weight 200 to 250 grams) in amounts sufficient to "significantly decrease" uptake of radioiodine by the fetuses (0.1 to 6.0 mg potassium perchlorate per adult rat). There was no evidence of embryo- or fetotoxicity at these doses. Study results reported by Postel (1957), Lampe *et al.* (1967) and Sztanyik and Turai (1988) suggest that perchlorate can pass through the placenta from the mother to the fetus in the test animals.

A developmental neurotoxicity study of ammonium perchlorate in rats was conducted by Argus Research Laboratories (1998a; 1998b; 1998c). Ammonium perchlorate was administered to groups of 25 female Sprague-Dawley rats via drinking water at target doses of 0, 0.1, 1.0, 3.0, and 10 mg/kg-day. The dosing period was from the beginning of gestation (DG 0) to post-natal day 22 (PND22). Five dams per group were selected for sacrifice and blood collection on PND10. Pups (F1 generation) were counted and clinical signs recorded daily during pre- and postweaning. Some of the pups were assigned to four different subsets for additional evaluations: Subset 1 for brain weight and neurohistological examination on PND12; Subset 2 for passive avoidance testing, water maze testing, and finally blood collection for thyroid and pituitary hormone analysis; Subset 3 for motor activity evaluation and auditory startle habituation; Subset 4 for

regional brain weight evaluation and neurohistological examination on PNDs 82 to 85. U.S. EPA (2002) analyzed the F1 data and concluded that perchlorate treatment was associated with: (a) brain morphometric changes in the 10 mg/kg-day dose group and possibility also the 3 mg/kg-day dose group; (b) thyroid colloid depletion, hypertrophy, and hyperplasia in the 0.1 and 3 mg/kg-day dose groups; (c) thyroid hormone (T3 and T4) changes in the 0.1 and 1 mg/kg-day dose groups; and (d) increases in motor activity in some dosed animals.

Argus Research Laboratories (1999; York *et al.*, 2001) reported a two-generation reproductive toxicity study in Sprague-Dawley rats. Male and female rats (30 rats/sex/group) of the first generation (P) were exposed to ammonium perchlorate in drinking water at 0, 0.3, 3, and 30 mg/kg-day. One male and one female were allowed a cohabitation period of a maximum of 14 days. Day 1 of lactation (LD1, postpartum) was defined as the day of birth. Rats that did not deliver a litter were sacrificed on gestation day 25 and examined for pregnancy status. At the end of the 21-day postpartum period, all surviving P1 rats were sacrificed. Pups not selected for continued evaluation were also sacrificed on LD21. The selected F1 pups were dosed during the postweaning, cohabitation, and lactation periods. All F1 generation dams and their litters (F2 generation) were sacrificed on LD21. York *et al.* (2001) reported that perchlorate is not a reproductive toxicant in the doses tested. In both the P and F1 adult rats, there were no deaths, abortions, or premature deliveries. No changes were reported in any sperm parameters in either P or F1 adult male rats nor on mating or fertility parameters in either P or F1 adult female rats (estrous cyclicity, fertility index, number of days in cohabitation, and number of rats mated). Natural delivery and litter observations for both F1 and F2 generation pups were comparable among the treated and control groups. Treatment-related effects were not observed on the gestation index, the number of dams delivering litters, the duration of gestation, the average number of implantations, the average number of live pups, the viability and lactation indices, the sex ratios, or the pup body weights.

York *et al.* (2001) found that perchlorate exposure caused statistically significant, dose-dependent changes in thyroid weight, histopathology, and hormone levels in P, F1, and F2 generation rats. Relative thyroid weights were significantly increased in the 30 mg/kg-day dose group for both sexes in the P generation and for F2 generation pups. However, in the F1 generation adult rats, relative thyroid weights were significantly increased in all dose groups for females and in the 3 and 30 mg/kg-day dose groups for male rats. All three generations developed hypertrophy and hyperplasia of thyroid follicular epithelium that increased in incidence and severity in a dose-related manner. Dose-related changes in TSH, T3, and T4 were also observed in the treated rats. However, these changes were inconsistent among the different generations, sexes, and ages of animals.

U.S. EPA (2002) noted that two male rats from the high dose group (30 mg/kg-day) in the F1 generation (second parental generation) in the study had adenomas of the thyroid. These males were dosed from conception to 19 weeks of age. Without incorporating historical data, the difference between 0/30 in the control and 2/30 in the 30 mg/kg-day is non-significant by standard tests (e.g., Fisher's exact). However, using two earlier reported background incidence rates of 3.6 percent and 3.9 percent for thyroid follicular

cell adenomas in male Sprague-Dawley rats in 2-year studies and Bayesian analysis, U.S. EPA (2002) determined the increase in thyroid follicular cell adenoma at 19 weeks in male Sprague-Dawley rats exposed to 30 mg/kg-day to be significant.

Effects of perchlorate on motor activity in Sprague-Dawley rats were studied by Bekkedal *et al.* (2000)). The females were dosed with ammonium perchlorate in drinking water for two weeks at 0, 0.1, 1, 3, or 10 mg/kg-day prior to mating with the breeder males and through PND10. As dosing was stopped on PND 10, it is likely that the pups were not directly exposed to perchlorate in drinking water. On PND14, one male and one female were randomly selected from each litter to be used in the motor activity testing. These same animals were tested on PND14, PND18, and PND22. Pups were individually tested in automated Opto-Varimex Activity boxes where 9 different measures of activity were recorded for 90 minutes on each test day. Data were analyzed in 9, 10-minute blocks using a repeated measures ANOVA.

Bekkedal *et al.* (2000) reported no statistically significant differences for any of the 9 measures of motor activity, and there were no reliable interactions related to treatment. A general pattern in the results was noticed. The authors suggested that there was a divergence in activity between the control and treated groups which emerged late in the 90-minute testing sessions.

U.S. EPA and NIEHS used a Bayesian hierarchical model to analyze the motor activity data reported by Argus Research Laboratories (1998a) and Bekkedal *et al.* (2000). They built a linear mixed-effects regression model relating dose, sex, age, habituation time and a habituation time \times dose interaction term to the expected number of ambulatory movements, with an animal-specific intercept included to account for within-animal dependency (U.S. EPA, 2002). U.S. EPA concluded that there was evidence of an increasing dose-response trend in motor activity in both data sets, and suggested that the lower limit on the estimated dose corresponding to a 10 percent increase in motor activity relative to control can be used as a surrogate for the NOAEL. Because of the variability in the Argus Research Laboratories (1998a) study, a NOAEL that relied on the Bekkedal *et al.* (2000) study was chosen at 1 mg/kg-day to represent effects on motor activity from these combined data.

Argus Research Laboratories (2001) studied the effects of perchlorate on thyroid and brain development both during gestation and postnatally. Perchlorate was administered in drinking water to female rats two weeks prior to cohabitation at 0, 0.01, 0.1, 1, or 30 mg/kg-day and continued through the day of sacrifice. F1 generation rats were not directly dosed but might have been exposed in utero during gestation and via maternal milk and maternal water during the postpartum period. The rats were selected only from female rats that had litters of at least 12 live offspring at the time of Cesarean-sectioning (Part A) or at the time of the first tissue collection (Parts B and C). P generation rats assigned to Parts B and C that delivered a litter were sacrificed on either PND 9 (Part B) or PND 21 (Part C). The thyroid and brain from one male and one female pup per litter were selected for histological and morphometric evaluation, with one set evaluated on PND4, PND9, and PND21. Details of the study and findings are described in the study reported prepared by the Argus Research Laboratories (2001), and a summary of the findings and evaluations are provided below.

According to the report, there were no deaths, adverse clinical observations or necropsy findings during the premating, gestational and/or lactation periods that were treatment-related in Parts A, B, and C. There were 16 pregnant dams in Part A. No treatment-related changes were found in Caesarean-sectioning or litter parameters. There were 15 or 16 pregnant dams for Parts B and C that delivered. Natural delivery was unaffected by the treatment and all clinical and necropsy observations in the F1 generation pups were considered unrelated to the treatment.

The absolute and relative thyroid weights of dams at the highest dose were increased. The absolute thyroid weights of some pups exposed at 1 and 30 mg/kg-day were increased. Furthermore, the absolute thyroid weights of the PND9 male pups in the 0.01, 0.1, 1 and 30 mg/kg-day dose groups were significantly increased over the controls.

An exposure-related increase in the incidence and severity of decreased colloid was noted in dams in the 1 or 30 mg/kg-day groups. Similar observations were also made on the fetuses at birth and pups at PND4 and PND9. An increased incidence of follicular cell hypertrophy and/or hyperplasia was found in dams in the 30 mg/kg-day dose group. An increased incidence of follicular cell hyperplasia was also found in the 1 mg/kg-day dams sacrificed on PND21.

In Part A, maternal TSH levels were significantly increased and T4 levels were significantly decreased at all exposure levels. Fetal TSH levels were significantly increased at 1 and 30 mg/kg-day while T3 was significantly decreased at all exposure levels. Changes of both the maternal and fetal hormone levels occurred in an exposure-dependent manner.

Maternal and fetal thyroid and pituitary hormone levels were also affected by various doses of perchlorate in Parts B and C. Most changes occurred in an exposure-dependent manner. In the PND21 male pups, TSH levels were significantly increased and T4 levels were significantly decreased at all exposure levels. T3 levels were also significantly decreased in the 1 and 30 mg/kg-day groups. In the PND21 female pups, TSH levels were increased at all exposure levels, reaching significance in the 0.1, 1, and 30 mg/kg-day groups. T4 levels were decreased with increased exposure but did not reach statistical significance.

Size of various brain areas was also measured in brain sections from the PND9 and PND21 pups. Due to signs of disruption or damage found in the PND9 sections that might have compromised the measurements, U.S. EPA (2002) relied upon the PND21 measurements. In the PND21 brains, the striatum, cerebellum, and corpus callosum all showed significant differences from control with the lowest administered dose of ammonium perchlorate, 0.01 mg/kg-day. As shown in Figure 3, different brain regions show an inverted U or U-shape dose response. For instance, the corpus callosum showed a notable increase in linear extent of 24 percent or more at PND21 in the 0.01, 0.1, and 1 mg/kg-day dose groups; however, this effect was not observed at the highest dose group, 30 mg/kg-day. Using this data, U.S. EPA (2002) identified a LOAEL of 0.01 mg/kg-day for the adverse effects of ammonium perchlorate on the developing brain in rats. This is equivalent to 0.0085 mg/kg-day for the perchlorate anion alone.

The design and implementation of this study have been criticized, and U.S. EPA's interpretation of the study data has been challenged (TERA, 2003). It was noted that the

way the brain sections were prepared and the method used to measure different regions of the brain are susceptible to experimental artifacts. The study has no positive control and it is not clear if the observed neurodevelopmental changes are related to thyroid hormone disruption. The association of some of the brain measurements with perchlorate exposure has also been questioned as there are no clear dose-response relationships. There are also concerns about the statistical method used in the U.S. EPA analysis.

Thuett *et al.* (2002) studied the effects of *in utero* and lactational exposure to ammonium perchlorate on developing deer mice. Breeding pairs were dose continuously with 0, 1 nM, 1 μ M, or 1 mM of ammonium perchlorate in drinking water, from cohabitation until pups were sacrificed at postnatal day (PND) 21. Pups from the second litter were used for evaluation. The researchers found the treated groups tended to have smaller litter sizes than did controls, but a greater survival percentage. They reported that perchlorate is a developmental toxicant and showed variable effects with increasing concentrations. Body weights of the pups in the 1 μ M group were consistently lower than in the controls and in other treatments after PND 1. They also reported that the treatment had an effect on the liver and heart weights. However, although liver weight alone was significantly different between treatments, liver weight when analyzed with body weight as a covariate showed no significant differences. Heart weights for male pups were decreased in the 1 μ M and 1 mM treatment groups. Heart weights decreased while body weight was increasing. Citing other study results, Thuett *et al.* (2002) suggested that an inadequate level of thyroid hormones during cardiac muscle development can alter cardiac function and/or heart size.

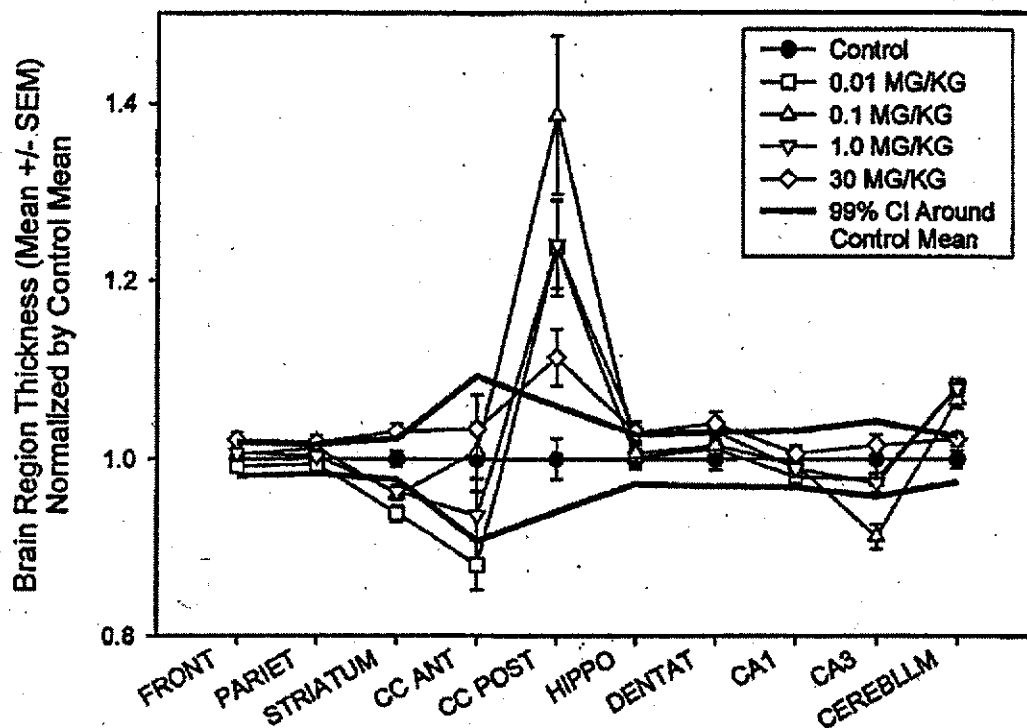


Figure 3. Profile analysis of brain morphometry measurements for PND21 rat pup brain regions.

The male and female data on linear thickness measurements were combined and normalized by the control mean of each region. The control data are represented by the horizontal line at 1.0. Profile analysis determines whether the vectors of measurements from each treatment group differ from each other and control in a dose-dependent fashion. The heavy line represents the ± 99 percent confidence interval around the mean control values. Note that while this plot uses the normalized data to more easily illustrate the data vectors, the actual analysis was performed using raw data values (from U.S. EPA, 2002).

Reproductive toxicity

Female rats were dosed with perchlorate in drinking water during gestation. The daily intake rates were estimated to range from 237 mg/rat to 615 mg/rat (Brown-Grant, 1966; Brown-Grant and Sherwood, 1971, as cited in U.S. EPA, 2002). These researchers observed no significant differences in litter size, number of pups, and pregnancy rate. Relative thyroid weights of the dams and litters were increased significantly compared with the controls.

A developmental toxicity study was performed on New Zealand White rabbits (Argus Research Laboratories, 1998d). It involved 25 naturally-mated does per group exposed to ammonium perchlorate in drinking water at 0, 0.1, 1.0, 10, 30, and 100 mg/kg-day from gestation day 6 to gestation day 28. Observations were based on 22, 24, 23, 24, 24, and 23 pregnant does that survived to gestation day 29 in the 0, 0.1, 1, 10, 30, and 100 mg/kg-day dosage groups, respectively. Fetuses were delivered by Caesarean section. The authors reported that doses as high as 100 mg/kg-day did not affect litter parameters. All values were within the historical ranges of the testing facility. The litter averages for corpora lutea, implantations, litter sizes, live and dead fetuses, early and late resorptions, percent dead or resorbed conceptuses, percent male fetuses and fetal body weights were comparable and did not differ significantly in the six dosage groups. All placentae appeared normal and no doe had a litter consisting of only resorbed conceptuses (Argus Research Laboratories, 1998d). U.S. EPA (2002) analyzed the maternal hormone data and noted statistically significant decreases in T4 for the 1, 10, 30, and 100 mg/kg-day dose groups. There were no statistically significant changes at any dose of T3 or TSH.

Argus Research Laboratories (1998d) also reported that no fetal alterations (defined as malformations and variations) were attributable to exposure to ammonium perchlorate at doses as high as 100 mg/kg-day: (a) the incidences were not dosage-dependent; (b) the observation occurred in only one or two high dosage group fetuses; or (c) the incidences were within the averages observed historically at the testing facility.

OEHHA notes that rabbit is probably not an appropriate animal model for the study of adverse developmental effects of perchlorate. Studies have shown that the placental iodide transport in rabbit is capable of generating a fetal serum-to-maternal serum iodide concentration of 5/1 to 9/1, thus facilitating the production of fetal thyroid hormones. A similar transport mechanism is not known to exist in human placenta (Hall and Myant, 1956 and Roti *et al.*, 1983, as cited in Fisher, 1996).

In a study by Argus (2000), female rats were dosed at 0, 0.01, 0.1, 1.0 and 30.0 mg/kg-day ammonium perchlorate in drinking water beginning 15 days before cohabitation and continuing through the day of sacrifice. All rats were sacrificed on gestation day 21, and a gross necropsy of the thoracic, abdominal, and pelvic viscera was performed. Preimplantation loss was noted at all dose levels: 12, 18, 20, 16, and 25 percent at the respective doses from 0 to 30 mg/kg-day. U.S. EPA (2002) noted that it was not clear whether the increase over control in this parameter was statistically or biologically significant. OEHHA analyzed the data by the Mann-Whitney U test (since the data are not normally distributed) and found that the increase in preimplantation loss was statistically significant in the 30 mg/kg-day group compared to controls ($p < 0.05$). A

decrease in the number of live fetuses was also reported to be statistically significant ($p < 0.05$) at 30 mg/kg-day, although no significant decrease was noted in the lower groups. Ossification sites per litter for sternal centers and forelimb phalanges were significantly reduced at 30 mg/kg-day.

Immunotoxicity

Shigan (1963) administered ammonium perchlorate to rabbits and white rats in water at 190 mg/kg-day for three months. The mode of administration was not described. No effect was found on immune function as evaluated by leukocyte phagocytosis (Shigan, 1963).

A series of hematological and immunotoxicology experiments in female B6C3F₁ or CBA/J Hsd mice was conducted as part of the U.S. EPA's perchlorate testing strategy (U.S. EPA, 1998a, 2002). In these experiments mice were exposed for 14 or 90 days to ammonium perchlorate at doses between 0.02 and 50 mg/kg-day via drinking water. The mice were tested at intervals for immunotoxicological effects such as delayed type hypersensitivity and cytotoxic lymphocyte activity (Keil *et al.*, 1998, 1999; Burleson Research Technologies).

In the hematological studies, there were no differences observed between control and dosed mice in 14- or 90-day experiments for erythrocyte cell count, hemoglobin, hematocrit, mean corpuscular volume, mean corpuscular hemoglobin, and mean corpuscular hemoglobin concentration; nor in leukocyte differential counts of neutrophils, monocytes, and lymphocytes. An increase in the percentage of reticulocytes was observed in the peripheral blood of mice exposed to 3 mg/kg-day of ammonium perchlorate in a 90-day study. No consistent alteration in the bone marrow stem cell assay was observed. An increase in the number of colony-forming units was observed in bone marrow cell cultures from mice dosed at 30 mg/kg-day in a 14-day study. However, in two other 90-day studies, this positive result was not confirmed. Upon reviewing the immunotoxicological studies, U.S. EPA (2002) found three immune function parameters that were altered by ammonium perchlorate exposure: (a) suppression of *in vitro* peritoneal macrophage phagocytosis of *L. monocytogenes*; (b) enhancement of the plaque-forming cell (PFC) assay response to sheep red blood cells (SRBCs), and (c) enhancement of the local lymph node assay (LLNA) to 2,4-dinitrochlorobenzene (DNCB).

Decreased *in vitro* phagocytosis of *L. monocytogenes* by peritoneal macrophages obtained from mice dosed for 14 days at 1 or 30 mg/kg-day (ammonium perchlorate) was observed. In mice exposed for 90-days, phagocytosis was decreased in all dose groups (Keil *et al.*, 1998, 1999). However, in a 90-day perchlorate exposure followed by a 30-day recovery period study, similar effects were not observed. These *in vitro* data suggest that perchlorate suppresses the phagocytic capacity of peritoneal macrophages, but this suppression is reversed after a 30-day recovery period. It is difficult to interpret the biological significance of this data set because *in vivo* study results indicate ammonium perchlorate exposure did not alter the ability of mice to combat *L. monocytogenes* infection. It was suggested that while perchlorate may reduce the phagocytic capacity of

peritoneal macrophages, the ability of macrophages from other sites (e.g., spleen, liver) to clear *L. monocytogenes* was not altered (U.S. EPA, 2002).

The PFC assay is routinely used for identifying immunosuppressive chemicals. The reason why the highest dose(s) of ammonium perchlorate, given over 90 days, enhanced this response is not known. The ELISA data for mice exposed to up to 30 mg/kg-day for 14 or 90 days do not corroborate this enhanced response to SRBCs as determined by the PFC assay. The data from Burleson Research Technologies (2000) indicate that exposure to perchlorate enhances the LLNA response to DNCB. While a dose of 50 mg/kg-day for 14 days enhanced the response, exposure to the same dose for 90 days suppressed the response. Exposure to lower doses of 0.06 and 0.2 mg/kg-day also increased the response; however, interpretation of these data is made difficult by the observation that exposure to 2 mg/kg-day did not affect the response, in the 14-day study. OEHHHA agrees with U.S. EPA (2002) that interpretation of the results is made difficult by (a) some technical problems encountered in the studies, (b) the apparent inconsistency of the high-dose study results, and (c) the unknown biological significance of the response enhancement.

Neurotoxicity

By interfering with the thyroid-pituitary axis, perchlorate interferes with development of the central nervous system. Thyroid hormone plays an essential role in the development of the corpus callosum and other brain structures. As part of the U.S. EPA's program to evaluate the toxicity of perchlorate, neurodevelopmental tests on Sprague-Dawley rats were conducted by Argus Research Laboratories (1998a, 2001). The study design and findings have been summarized in the "Developmental and Reproductive Toxicity" section.

Endocrine Toxicity

Many oral and injection studies documented the effects of perchlorate on the thyroid and pituitary hormones as well as the thyroid of the treated animals. The designs and findings of these studies are summarized in the "Subchronic Toxicity" and "Developmental and Reproductive Toxicity" sections.

Yu *et al.* (2000), working with the United States Air Force and U.S. EPA, investigated the inhibitory effects of perchlorate on thyroidal iodide uptake in rats. They injected perchlorate at 0, 0.01, 0.1, 1 or 3 mg/kg to groups of male Sprague-Dawley rats (6 animals per dose and time point). At 2 hr post dosing, the rats were challenged with ¹²⁵I with carrier (33 µg/kg) by intravenous injection and euthanized at various time points post dosing. Statistically significant thyroidal iodide uptake inhibition was found in the 1 and 3 mg/kg perchlorate dose groups at 2, 6, and 9 hr time points. In addition, significant inhibition was also observed in the 0.1 mg/kg dose group at the 9 hr time point (Table 6).

Table 6. Percent Inhibition of Iodide Uptake in the Thyroid Gland of Male Rats (n=6) Dosed with Perchlorate (Yu *et al.*, 2000)

Time points	Perchlorate dose (mg/kg)	Mean iodide concentration in the thyroid (µg/g)	Thyroidal iodide uptake inhibition * (%)
2 hours	Control **	24.4	-
	0.01	21.3	13
	0.1	18.6	24
	1	7.4 ^a	70
	3	3.0 ^a	88
6 hours	Control **	46.5	-
	0.01	36.7	21
	0.1	32.0	31
	1	19.2 ^a	59
	3	9.1 ^a	80
9 hours	Control **	55.0	-
	0.01	49.2	11
	0.1 ^a	39.2	29
	1	24.7 ^a	55
	3	10.0 ^a	82

* Percent inhibition = (control mean – dose mean) x 100 / (control mean)

** Dosed with ¹²⁵I with carrier only (33 µg/kg)

^a Significantly different from control at p<0.05

In a follow-up study, Yu *et al.* (2000) exposed groups of male Sprague-Dawley rats (6 animals per dose and exposure duration) to perchlorate in drinking water with target concentrations of 0, 1, 3, and 10 mg/kg-day continually for 1, 5, or 14 days. At the end of day 1, 5, or 14, rats were challenged once with 33 µg/kg ¹²⁵I with carrier and euthanized 2 hr later. Blood and thyroid gland were collected for analyses. A dose-related inhibition was noted in the one-day treatment group. The degree of inhibition was reduced over time and by exposure day 14, no inhibitory effect was observed in the 1 and 3 mg/kg-day groups. In a similar study, thyroid hormone profile of rats exposed to perchlorate was investigated. Male rats in groups of 8 were exposed to perchlorate in drinking water at 0, 0.1, 1, 3, and 10 mg/kg-day continually for 1, 5, or 14 days. In all treated groups, regardless of dose or exposure duration, TSH levels were increased compared to the control. The serum T4 levels were initially decreased in all dose groups except the lowest, 0.1 mg/kg-day. By 14 days, the 1 mg/kg-day dose group returned to control T4 values while T4 levels of the 3 and 10 mg/kg-day dose groups were still significantly depressed. Yu *et al.* (2000) suggested that the regulations of thyroidal

iodide uptake and serum T4 are rapid in rats and can compensate for the anti-thyroid actions of perchlorate at low doses.

Yu *et al.* (2002) modeled the effects of perchlorate on the hypothalamus-pituitary-thyroid axis in the male rat. They found a high correlation between serum concentrations of perchlorate and percentage inhibition in thyroidal iodide uptake, irrespective of the route of administration. They found the hypothalamus-pituitary-thyroid axis responded quickly to perchlorate blocking effects on thyroidal iodide uptake. Serum thyroid hormone levels decreased and serum TSH levels increased in response to perchlorate. Under the influence of TSH, the thyroid was up-regulated and was able to overcome the blocking effects of perchlorate by increasing its capacity to sequester iodide and produce hormones. Yu *et al.* (2002) noted that this is a dose-dependent phenomenon, which was overwhelmed by the blocking effects of high serum levels of perchlorate (corresponding to above approximately 1 mg/kg-day).

Clewell *et al.* (2001) developed and validated rat and human physiologically-based pharmacokinetic (PBPK) models for the estimation of the effect of life stage and species on perchlorate and iodide inhibition kinetics. In a later paper, Clewell *et al.* (2003) used the models to estimate perchlorate distribution in the male, pregnant, fetal, lactating and neonatal rat, and predict resulting inhibitory effects on thyroidal iodide uptake. They reported that the PBPK model predicted that while the lactating dam has the greatest internal perchlorate exposure, the fetal rat thyroid is most vulnerable to inhibition.

Carcinogenicity

A number of animal studies have been reported that may be useful in determining the carcinogenic potential of perchlorate. However, the interpretation of the study results is hampered by the small number of animals per dose group, short exposure and observation durations, lack of multiple dose groups, and co-exposure to other cancer causing agents.

Gauss (1972) treated female NMRI mice with one percent potassium perchlorate in the diet or the control diet for 160 days. The one percent dose is equivalent to approximately 2,000 mg/kg-day based on standard assumptions. The investigator noted progressive changes in the thyroids of treated mice beginning with colloid loss, progressing to increases in size of nuclei and increased epithelial height, followed by appearance of hyperplasia and hypertrophy of the thyroid parenchyma. Later in the treatment period, hyperplastic follicles, areas of adenomatous tissue, adenoma complexes and secreting cystadenomas were observed. No progression to malignancy was observed during the study period.

Several Japanese investigators (Hiasa *et al.*, 1987) tested potassium perchlorate for its ability to promote the carcinogenic activity of N-bis(2-hydroxypropyl)nitrosamine (DHPN). They divided the rats into four groups. Groups 1, 2, and 3 received 1000 ppm potassium perchlorate, 1000 ppm potassium iodide, or 1000 ppm propylthiouracil in the diet, respectively. Group 4 was the control and received the basal diet throughout the study period of 18 weeks. At the beginning of the study, 50 percent of the rats in each group were injected with DHPN at 280 mg/100 g body weight as an initiator. Rats treated with both chemicals had a 100 percent incidence (20/20) of thyroid adenomas.

Rats treated with potassium perchlorate alone had no thyroid adenomas. The incidence was five percent (1/20) in rats given DHPN alone. The investigators concluded that potassium perchlorate promoted the development of thyroid tumors in the rats treated with DHPN, but the chemical itself was not carcinogenic in this experiment.

Groups of male Wistar rats were exposed for two years to potassium perchlorate in drinking water at concentrations of 0 or one percent (Kessler and Kruskemper, 1966). Based on body weights and estimated water consumption, the one percent concentration was estimated to provide a dose of approximately 1,300 mg/kg-day. Animals were sacrificed and examined at intervals of 0, 40, 120, 220 and 730 days of exposure. Body weights of control and exposed animals were similar throughout the experiment, but thyroid weights of the exposed rats increased markedly compared to control rats at each interval of examination. At 40 days, the exposed rats developed follicular cell hyperplasia, i.e., small follicles with high epithelia, large nuclei, numerous mitoses, colloid resorption and low-grade mesenchymal reaction. According to the authors, these changes are typical of thyroid glands stimulated by TSH for a relatively short time. Diffuse degenerative changes with fibrosis and increased colloid were observed after 200 days. Four of 11 rats treated with one percent potassium perchlorate for two years developed benign thyroid tumors. The twenty untreated controls had no thyroid tumors.

Pajer and Kalisnik (1991) divided 72 female BALB/c mice into 6 groups. Three groups were treated with 1.2 percent sodium perchlorate, while three groups were controls. Eight or 32 weeks after the beginning of the study, one perchlorate and one control group of animals were irradiated with a total of 4 Gy of ionizing radiation (gamma rays) over a period of five days. Forty-six weeks after the beginning of the experiment, 42 animals were sacrificed, while 30 had died during the experiment. The perchlorate dose to the treated mice is approximately 2,100 mg/kg-day based on standard default assumptions about body weight and water consumption. Perchlorate treatment alone caused hypothyroidism with hypertrophic and hyperplastic thyroid epithelial cells as well as pituitary thyrotropic cells. Perchlorate and irradiation together caused effects similar to those caused by perchlorate treatment alone. Follicular cell carcinomas of the thyroid gland were found after perchlorate treatment and after perchlorate with irradiation. The incidences of these carcinomas are shown in Table 7 below.

Table 7. Carcinoma Incidence in Female BALB/c Mice Treated with Sodium Perchlorate in Drinking Water (Pajer and Kalisnik, 1991)

Treatment group	Thyroid follicular cell carcinoma incidence
Control (tap water)	0/22
Perchlorate in water (1.2%)	5/6 *
Perchlorate (1.2%) + irradiation (4 Gy)	14/14 *

* Statistically significant, Fisher exact test, $p < 0.001$.

Gy = Gray, the SI unit for measuring absorbed ionizing radiation.

The data indicate perchlorate caused thyroid follicular cell carcinomas in the treated mice. The study result is limited by the small number of animals in the perchlorate-only treated group, the design of the study, and the deaths of over 40 percent of the mice before the end of the experiment.

In a two-generation reproductive toxicity study in rats (Argus Research Laboratory, 1998b), two out of 30 male Sprague-Dawley rats (P2) in the highest dose group (30 mg/kg-day) were found to have adenomas of the thyroid. No such tumors were found in the control group or the other dosed groups (0.3 mg/kg-day and 3 mg/kg-day). In the study, the male rats were exposed to ammonium perchlorate in drinking water from conception to 19 weeks of age. As thyroid follicular cell adenomas are relatively rare in male Sprague-Dawley rats (the background incidence of this tumor reported in the literature was only 3.6-3.9 percent), U.S. EPA (2002) determined the increase in tumor incidence to be treatment related.

In a number of subchronic perchlorate studies, increased thyroid follicular cell hypertrophy and hyperplasia were observed in some of the treated animals (Lampe *et al.*, 1967; Caldwell *et al.*, 1995; Springborn Laboratories, 1998; Argus Research Laboratories, 1998a, 1999, 2001; Keil *et al.*, 1998). Details of the study results have been presented. Summaries of these data sets are provided in sections on "Subchronic Toxicity" and "Developmental and Reproductive Toxicity." These data indicate that oral administration of perchlorate induces hyperplasia in the thyroid of rodents and if the exposures are lengthened, some of the lesions might progress to thyroid tumors.

Toxicological Effects in Humans

The major adverse health effects of perchlorate at low dosages are associated with disruption of thyroid hormone balance. These effects are similar to those caused by iodine deficiency. At high doses, perchlorate exposure is known to cause other adverse health effects such as blood disorder. Some of the adverse health effects of iodine deficiency are discussed in the "Sensitive Subpopulation" section under "Dose-Response Assessment: Noncarcinogenic Effects."

Acute Toxicity

The acute lethal oral dose of perchlorate for an adult human is estimated to be 15 g, or 214 mg/kg for a 70-kg person (Von Burg, 1995).

Subchronic Toxicity

Potassium perchlorate has been used to treat Graves' disease³ in humans, and most of the prior data on perchlorate effects on humans are in patients with this disease. Perchlorate

³ Graves' disease is an autoimmune disorder in which patients carry immunoglobulins in their blood that bind to the TSH receptors on thyroid cells and act like TSH to stimulate DNA synthesis and cell divisions leading to a hyperthyroid state.

inhibits the excessive synthesis and secretion of thyroid hormones by inhibiting the uptake of iodide into the thyroid and causes a discharge of accumulated iodide in the gland.

Godley and Stanbury (1954) report using potassium perchlorate to treat 24 patients with Graves' disease. Patients were treated with 600 to 1,200 mg/day for at least 11 weeks with a few as long as 52 weeks. Two patients developed gastrointestinal problems. In one patient, these effects occurred at 600 mg/day.

Crooks and Wayne (1960) administered potassium perchlorate at 600 to 1,000 mg/day to 200 patients with Graves' disease and observed one case of skin rash and three cases of nausea. In another group of 10 patients given 1,500 mg/day and 40 patients given 2,000 mg/day, five cases of skin rash, two cases of nausea, and one case of agranulocytosis occurred. Leukocyte counts returned to normal in the patient with the agranulocytosis when perchlorate treatment was stopped. The length of treatment was unclear but generally appears to have been less than 8 weeks; it appears that one patient was followed for 22 weeks.

Morgans and Trotter (1960) reported that 3 percent of 180 patients treated with 400 to 1,000 mg/day potassium perchlorate and 18 percent of 67 patients treated with 1,200 to 2,000 mg/day displayed a variety of adverse reactions that included skin rash, sore throat, gastrointestinal irritation, and lymphadenopathy. Based on the data reported by Crooks and Wayne (1960) and their own clinical observations, Morgans and Trotter (1960) recommended a daily dose of 800 mg/kg-day, a compromise between effectiveness and minimizing the toxic side effects of perchlorate.

Genetic Toxicity

No reports were found of studies that examined genetic endpoints (chromosomal aberrations, sister chromatid exchanges, etc.) in humans exposed to perchlorate.

Chronic Toxicity

Connell (1981) reported a case of a female Graves' disease patient who was treated with 200 mg/day perchlorate for 22 years with good control of the disease, and no apparent adverse effects.

Developmental and Reproductive Toxicity

Clinical Studies

Crooks and Wayne (1960) administered potassium perchlorate at 600 to 1,000 mg/day to a group of pregnant women that were suffering from hyperthyroidism and observed a very slightly enlarged thyroid in 1 of the 12 infants born to the mothers. They also reported that the enlarged thyroid returned to normal size in 6 weeks, and no other abnormalities were observed. Several key parameters were not provided in the paper: detailed dosage information, time of perchlorate treatment in relation to the gestation

period, thyroid function of the newborns, and the neurological as well as behavioral development of the offspring. Furthermore, interpretation of the result is made difficult by the fact that the women were suffering from thyrotoxicosis (excess quantities of thyroid hormones).

Ecological and Epidemiological Studies

A preliminary health review of a potentially exposed area of Rancho Cordova, CA by the California Department of Health Services (DHS, 1997) analyzed several state databases for possible perchlorate-related adverse health effects during the suspected years of drinking water contamination within the area or region most likely exposed. The analysis of newborn thyroid hormone data for the period 1985 through 1996 did not indicate a positive correlation between residence in the potentially-exposed areas and neonatal hypothyroidism. The TSH levels of neonates with initially low T4 levels in the potentially-exposed areas were found to be statistically lower than those in the control areas, contrary to what was expected.

Using similar data from a statewide newborn screening registry (California Newborn Screening data), Schwartz (2001) evaluated the serum T4 and TSH levels of all newborns in California during 1996. All infants were screened by their serum T4 levels; the samples of those with a low T4 (i.e., less than or equal to 9 mg/dL and the next lowest 5 percent of the values in each tray of samples) were further analyzed for TSH levels. Perchlorate exposure was estimated based on the following information: postal zip code, concentration of perchlorate measured in underground water sources sampled between February 1997 and June 2000, water source production, water purchases and sales, and characteristics of the water distribution system. After adjustment for gender, ethnicity, multiple birth, birth weight, and age at time of blood sample (in 6-hr increments), and separation of infants into groups whose mothers were presumably exposed to low (1-2 ppb), medium (3-12 ppb), and high (≥ 13 ppb) levels of perchlorate in their drinking water, the study found statistically significant associations between putative exposure and serum thyroid hormone levels (T4 and TSH) (Table 8). Substantial birth weight, gender, ethnicity, age at sampling, and birth multiplicity effects were observed for T4, and smaller effects for TSH. The author noted that 90 percent of the variability in T4 and TSH infant hormone levels remained unexplained after controlling for all the parameters described. Two variables that were not controlled, gestational age and laboratory measurement variability, may explain some of the variability not already accounted for in the model. Among the ecological studies described in this section, this study controlled for most of the important variables and found that even an extremely low level of perchlorate exposure through drinking water consumption was associated with a decreased serum T4 in newborns in California. There are uncertainties associated with the exposure estimates and this difficulty is alleviated to a degree by categorizing the infants into three dose groups (low, medium, and high). Another source of uncertainty is that the serum T4 and TSH data were from 1996, but the perchlorate in water source data were collected later, between February 1997 and June 2000.

Kelsh *et al.* (2003) criticized the TSH results because more than 50 percent of the data had the value recorded as the minimum, 5 μ U/mL. Thus the true variance of TSH is reduced, which inflates the level of significance.

Table 8. Perchlorate Exposure and Thyroid Function in California Neonates; Infant Characteristics versus Perchlorate Levels in Drinking Water (Schwartz, 2001)

	Perchlorate concentration				Test statistic p value (df)
	None (0 ppb)	Low (1-2 ppb)	Medium (3-12 ppb)	High (\geq 13 ppb)	
Number of infants (%)	255,382 (49.5)	127,041 (24.6)	131,483 (25.5)	1,945 (0.4)	
Female, %	48.8	48.8	49	49	
T4, mg/dL Mean (SD)	170.9 (50.9)	162.1 (48.7)	160.6 (48.3)	150.5 (44.0)	F=1,649.6 p<0.0001 (3)
TSH, μ U/mL Mean (SD)	7.6 (18.8)	7.6 (19.8)	7.7 (19.4)	7.9 (4.6)	F=0.05 p=0.9 (3)

SD = standard deviation

df = degree of freedom

Kelsh *et al.* (2003) used the California Newborn Screening database to study the thyroid health of newborns whose mother resided in Redlands during the period 1983 through 1997. Perchlorate has been detected in groundwater wells in Redlands. Two parameters, primary congenital hypothyroidism and elevated concentrations of TSH, were used to represent the thyroid health. Newborns of San Bernardino and Riverside counties excluding newborns from Redlands and communities where perchlorate has been detected were used as the comparison group. All residents in the Redlands community were assumed to be uniformly exposed to perchlorate. However, there is a lack of information about perchlorate levels in drinking water of the study area over the 15-year study period. The detection limit of perchlorate in water before 1997 was about 400 ppb. In the 2001 and 2002 Consumer Confidence Reports, the City of Redlands reported that the concentrations of perchlorate in its water system ranged from non-detect to 9 ppb, with an average concentration below 1 ppb.

Kelsh *et al.* found no excess prevalence of primary congenital hypothyroidism in Redlands newborns over the 15-year study period. They reported an adjusted standardized prevalence ratio of 0.45 and a 95 percent confidence limit of 0.06-1.64. Kelsh *et al.* also did not find evidence of elevated TSH for Redlands newborns either using San Bernardino and Riverside (excluding Redlands) newborns or using only San Bernardino newborns as the comparison group. The Colorado River is one of the water sources of Riverside, so the water serving this community, the "unexposed group", is also likely to have been contaminated with perchlorate. The researchers found that Hispanic ethnicity, low and high birth weights, and female sex are risk factors for primary

congenital hypothyroidism. High TSH levels are mostly influenced by the age (in hours) when the blood samples are collected. Immediately after birth, a normal surge in TSH occurs so that early collection of blood samples (<24 hours) can produce high TSH results. The study results are limited by the small number of observations for the Redlands community, due to the rarity of primary congenital hypothyroidism and elevated TSH status. Like many other ecological studies, this study is also limited by the lack of detailed exposure information for the residents of Redlands.

Crump *et al.* (2000) studied a total of 162 school-age children (6-8 years old) and 9,784 newborns in three proximate cities in northern Chile that have different concentrations of perchlorate in drinking water: Taltal (100 to 120 µg/L), Chañaral (5-7 µg/L), and Antofagasta (non-detectable: <4 µg/L). Approximately 25 separate water sources were sampled in each study city. Water samples were taken from portable water faucets at participating schools, homes of students, and public buildings located near the schools. Crump *et al.* (2000) reported that mean levels of TSH, T4, free T4, and T3 of the school-age children were very similar among the three cities. Among all the school-age children, there was a small, non-significant increased risk of goiter in Chañaral (26.5 percent) and Taltal (23.3 percent) compared with Antofagasta (17 percent). However, the high background goiter prevalence rate in Antofagasta represents an unexplained, confounding variable.

It should be noted that Crump *et al.* (2000) also reported relatively high urinary iodine excretion of the schoolchildren in all three cities. They reported mean iodine excretion levels of 76.6 µg/dL, 61.4 µg/dL, and 75.6 µg/dL for Taltal, Chañaral, and Antofagasta, respectively. These levels are much higher than those in the NHANES III database, which found a mean urinary iodine level of 30.5 µg/dL in the 6-11 years old age group of children in the U.S. The high urinary iodine levels and high goiter prevalence among the schoolchildren in the Chile study make the interpretation of thyroid function data difficult, and together they indicate there may be other confounding factors.

Crump *et al.* (2000) also found that schoolchildren with lifelong residence in Taltal were 5 times more likely to report a family history of thyroid disease compared with schoolchildren with lifelong residence in Antofagasta, adjusted for age, sex, and urinary iodine (Table 9). Chañaral children had no increased prevalence of self-reported family history of thyroid disease. Families of 19 out of 61 (31 percent) children in Taltal were reported as having some history of thyroid disease. Twelve of these families reported having a single relative (usually a mother or grandmother) with goiter, hypothyroidism, or unspecified thyroid disease; and seven reported having two or more relatives with thyroid disease. Reasons for these observations are yet unknown. Crump *et al.* (2000) speculated that the findings might reflect a change in exposure patterns over the past several generations. Because iodized salt was not introduced in the region until 1982, it is possible that a combination of low iodine intake and the relatively high perchlorate exposures in Taltal was sufficient to cause thyroid problems in past decades.

Assuming the level of perchlorate contamination of the city of Taltal has not changed significantly in the last few decades, a LOAEL of 100 µg/L for familial thyroid problems can be identified from the study reported by Crump *et al.* (2000). Applying the default values of 2 L/day for drinking water consumption and 70 kg for an adult body weight, the LOAEL is equivalent to 3 µg/kg-day. It must be emphasized that in addition to

perchlorate exposure, a low dietary iodine intake might have also contributed to the increased thyroid problems reported in the residents of Taltal.

Table 9. Odds Ratios for Association Between Self-Reported Family History of Thyroid Disease* Among Schoolchildren and City of Residence (from Crump *et al.*, 2000)**

City	Schoolchildren with less than lifelong residence (n=162)		Schoolchildren with lifelong residence (n=127)	
	Odds ratio	95% confidence interval	Odds ratio	95% confidence interval
Antofagasta	1.00	-	1.00	-
Chañaral	0.89	0.25-3.19	1.04	0.21-5.09
Taltal	3.35	1.19-9.38	4.97	1.29-19.17

* Direct relative (parent, sibling, grandparent, great-grandparent, aunt, uncle, or cousin) with history of goiter, hypothyroidism, or subtotal thyroidectomy.

** Adjusted for age, sex, and urinary iodine; excluded one child with autoimmune hypothyroidism.

No cases of congenital hypothyroidism were detected in cities with detectable perchlorate in drinking water (Taltal and Chañaral). Among newborns analyzed in the study, seven presumptive cases of congenital hypothyroidism ($TSH \geq 25 \mu U/mL$) were detected, all originating in Antofagasta, corresponding to an incidence rate in Antofagasta of 1 per 1270 newborns. The national average rate of congenital hypothyroidism in Chile between 1992 and 1999 is 1 per 3484 newborns screened. Maximum TSH levels observed in Chañaral and Taltal were $17.1 \mu U/mL$ and $13.5 \mu U/mL$, respectively. Adjusted for sex and age, linear regression comparisons of log TSH of the newborns by city showed that average log TSH in Taltal was significantly lower compared with those of the other two cities. However, for the group of newborns sampled on day 1-2, the mean and the median TSH levels of Taltal ($4.2 \pm 1.2 \mu U/mL$ and $4.2 \mu U/mL$) were higher than those of Chañaral ($3.2 \pm 3.5 \mu U/mL$ and $1.9 \mu U/mL$) and Antofagasta ($3.2 \pm 1.9 \mu U/mL$ and $2.7 \mu U/mL$). However, interpretation of these data is complicated by the high variability of TSH levels in newborns (1-2 days).

Low levels of perchlorate (4 to $16 \mu g/L$) have been detected in the drinking water supplies of six counties in California (Los Angeles, Orange, Riverside, Sacramento, San Bernardino, San Diego) and one in Nevada (Clark). Lamm and Doemland (1999) evaluated the congenital hypothyroidism incidence rates of the seven counties and compared them with the state rates. All infants were screened by their serum T4 levels; those with a low T4 (i.e., less than the 10 percentile) were further screened for high TSH levels. An infant is considered to be potentially congenitally hypothyroid if the serum $TSH \geq 25 \mu U/mL$. Infants that are judged positive on this test are diagnosed by a physician to confirm whether they are suffering from congenital hypothyroidism.

County-specific, ethnicity-specific data for the two states were obtained for 1996 and 1997. Within the seven counties, nearly 700,000 newborns were screened. In all, 249 cases were identified, where 243 were expected based on the state incidence rate, for an overall risk ratio of 1.0 (95 percent confidence interval, 0.9 to 1.2). The risk ratios for the individual counties ranged between 0.6 and 1.1. Based on these results, Lamm and Doemland (1999) concluded that the study did not indicate an increase in the incidence of congenital hypothyroidism with the reported perchlorate levels. Although Clark County of Nevada obtains nearly all its water from Lake Mead, which is known to be contaminated with perchlorate, the six California counties obtain their water from multiple sources and many of them are not contaminated with perchlorate. Because of this, there is a significant uncertainty in the estimation of the level of perchlorate exposure in the California counties. Nevertheless, it is noted that out of the 36,016 newborns screened in Clark County between 1996 and 1997, seven cases were identified, where 8.3 cases were expected. The risk ratio was 0.8, with 95 percent confidence limits of 0.34 to 1.74.

In a related study, Lamm and his colleagues compared serum T4 levels of newborns (1 to 4 days of life) from the city of Las Vegas, Clark County, Nevada, which has perchlorate in its drinking water, and those from the city of Reno, Nevada, which does not (detection limit, 4 µg/L) (Li *et al.*, 2000a). A total of 17,308 newborns from Las Vegas and 5,882 newborns from Reno during the period April 1998 through June 1999 were included in the study. Drinking water perchlorate levels measured monthly for Las Vegas during the study period ranged from non-detectable to 15 µg/L. Li *et al.* (2000a) reported that they observed no differences in the mean blood T4 levels (approximately 17 µg/dL) of the newborns from the two cities. It is important to note that variables, such as age at time of blood sampling, birth weight, and ethnicity, that were shown to be associated with serum T4 levels in the study reported by Schwartz (2001), were not controlled for in this study.

Based on the perchlorate concentration data of Las Vegas for the 9 months preceding the study period, Li *et al.* (2000a) also estimated the cumulative prenatal perchlorate exposure of the newborns. The Las Vegas newborns during the study period would have had maternal perchlorate exposures that ranged between 9 ppb-months and 83 ppb-months; the Reno newborns during this period were presumed to have had 0 maternal ppb-months. For each month, the difference between mean neonatal T4 levels of the two cities was examined, along with the cumulative maternal perchlorate exposure. Linear regression analysis showed no evidence of an association (slope = -0.0003; $R^2 = 0.002$). However, it is not clear how the cumulative exposures were calculated, especially the statistical treatment of non-detects. As discussed earlier, serum T4 and TSH levels of neonates may not be the most sensitive end-points or indicators of perchlorate exposure. For women with marginal iodine deficiency, additional stresses to the thyroid due to pregnancy and perchlorate exposure can cause low maternal T4, which may in turn impair the brain development of the fetus.

In another study, Li *et al.* (2000b) studied neonatal blood TSH levels sampled between December 1998 and October 1999 in Las Vegas (up to 15 ppb perchlorate in drinking water) and Reno (no perchlorate in drinking water). Serum TSH levels were determined on screening samples that were below the 10th percentile on T4 in each daily batch of samples. The study samples were from newborns in their first month of life (excluding

the first day of life, when TSH levels are unstable) with birth weights of 2.5 – 4.5 kg. The authors found neonatal TSH levels were not associated with perchlorate exposure of less than or equal to 15 ppb ($p=0.97$). However, lack of control of several variables, such as age at sampling, birth weight, and ethnic origin, that are known to affect serum T4 and TSH levels in infants makes interpretation of the results difficult.

Brechner *et al.* (2000) reported a study that showed an association between low-level perchlorate exposure and serum TSH levels in newborns. They compared serum TSH levels of newborns in Yuma, a city that obtains its public drinking water entirely from the Colorado River below Lake Mead, with newborns in Flagstaff, a city that obtains none of its public drinking water from the Colorado River below Lake Mead. Lake Mead is known to have perchlorate contamination. However, no useful water monitoring data were available for Yuma and Flagstaff during the study period (between 1994 and 1997). This is because the detection limit of perchlorate in water was 400 ppb during that period of time. Since March 1997, the detection limit of perchlorate has been improved to approximately 4 ppb. In August 1999, U.S. EPA reported that perchlorate levels in Yuma were 6 ppb in both raw water and finished drinking water (Brechner *et al.*, 2000).

Brechner *et al.* (2000) found that median newborn TSH levels in Yuma were significantly higher than in Flagstaff. According to Brechner *et al.*, this remained true after adjusting for factors known or suspected to affect serum TSH levels in newborns, including age in days at measurement of the first TSH level and race/ethnicity. However, the major factor influencing the TSH levels was the time (after birth) at which blood samples were collected, which was significantly different in the two cities. In addition, only about 10 percent of all the blood samples were analyzed for TSH and the selection procedure was not random; the statistical biases thus introduced might have affected the results.

As described above, several published and unpublished ecological studies report investigations of the relationship between perchlorate in drinking water and thyroid function in newborns and children. There are several challenges regarding the interpretation of these study results. First, there are considerable uncertainties in the estimation of perchlorate exposure in all the studies. The level of contamination of an area or region was often estimated based on only a few samples. It was assumed that the concentration of perchlorate in the water in an area or region could serve as a proxy for individual exposure. Second, some confounding factors were either not controlled or only partially controlled for in the studies. For example, in the analyses of thyroid hormone levels in Las Vegas and Reno newborns (Li *et al.*, 2000b), ethnic differences were not accounted for. In studies reported by DHS (1997) and Brechner *et al.* (2000), the control on the age at sampling was not adequate. As serum T4 and TSH levels are known to change significantly during the first 24 hr after birth, knowing the age at sampling (in hours) is very important in this type of study. Finally, small sample size and relatively low prevalence rate of the event of interest might have severely limited the sensitivity of the studies.

There is a concern about the way the neonate blood samples were selected for TSH determination in the studies reported by Schwartz (2001), Brechner *et al.* (2000), and Li *et al.* (2000b). In all these studies, TSH levels were only determined on screening samples that were below a certain cut-off value or below a certain (e.g., 10th) percentile on T4 in each daily batch of samples collected. Bias might have been introduced by this

sampling procedure, as the selection was not random and serum T4 level is not well correlated with serum TSH level, especially during the first 48 hr of life (Figure 4).

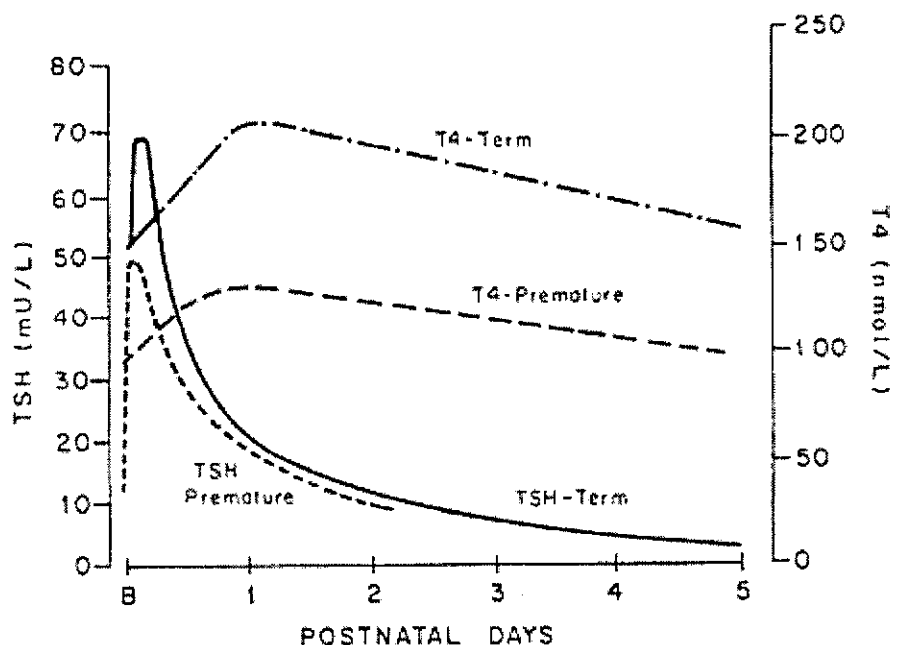


Figure 4. Changes in serum TSH and T4 concentrations in full-term and premature infants during the first five days of life. The neonatal TSH surge peaks at 30 minutes in both preterm and term infants and is followed by a progressive increase in serum T4 concentrations, peaking at 24 to 36 hours (from Fisher and Klein, 1981).

Immunotoxicity

Weetman *et al.* (1984) investigated the effect of perchlorate on human T and B cell responses to mitogen *in vitro*. Perchlorate at concentrations of 0, 0.01, 0.1, 1.0 and 10 mmol/L (1.17 g/L) were tested in cultures "designed to assess B and T cell responses." Supernatant IgG and IgM were measured by enzyme-linked immunoassays after culture for 10 days with pokeweed mitogen. The investigators found that perchlorate at 0.1 to 10 mmol/L inhibited IgG production and at 10 mmol/L inhibited IgM production. They concluded that perchlorate has significant immunosuppressive activity at pharmacologically relevant concentrations that is not due to simple cytotoxicity (assessed by ethidium bromide/acridine orange fluorescence).

Endocrine Toxicity

Clinical Studies

Stanbury and Wyngaarden (1952) studied the effect of perchlorate on the discharge and uptake of iodide by the thyroid in Graves' disease patients. To study the effect of perchlorate on the discharge of accumulated iodine, they gave 30 mg of 1-methyl-2-mercaptoimidazole orally to eight patients. A dose of 200 mg of propylthiouracil was given to a ninth patient. One hour later, a tracer of I^{131} was given. The accumulation of this isotope in the neck was recorded at frequent intervals until it was leveling off. At this point, quantities of potassium perchlorate varying from 3 to 500 mg were given orally in small volumes of water. In each patient except the one treated with propylthiouracil there was a sharp fall in the counting rate within a few minutes after the ingestion. This always occurred within 30 minutes. With smaller doses the discharge of the I^{131} was incomplete, but doses of 100 mg caused a fall in counting rates nearly to the counting rates recorded from the thigh (background). The investigators also reported that a single oral dose of 10 mg perchlorate caused about a 50 percent release of accumulated iodine. Potassium perchlorate doses as low as 3 mg (equivalent to 2.2 mg perchlorate) caused detectable, but incomplete, release of iodide from the thyroid. Assuming an adult body weight of 70 kg, it is equivalent to an oral dose of 31 $\mu\text{g/kg}$.

A LOAEL is not identified for this experiment because of the following reasons: (a) the number of patients per dose group is not known, (b) acute exposure, (c) the patients suffered from a thyroid disease which might have affected iodide uptake, and (d) the patients were pretreated with drugs (either 1-methyl-2-mercaptoimidazole or propylthiouracil) that may enhance the release of iodide in the thyroid gland by preventing the oxidation of iodide ion to iodine and thyroid hormone synthesis.

To study the effect of perchlorate on the uptake of iodide in the unblocked gland, Stanbury and Wyngaarden (1952) gave 100 mg of potassium perchlorate to three patients and an hour later, tracers of I^{131} . No thyroid hormone-disrupting drugs were given. Several days later each patient received a control tracer without previous perchlorate. In two cases, the studies were continued for 48 hours, but in the third a period of observation of only five hours was possible after the tracer. For the two patients with long observation time, the control uptake was about 70 percent of the administered dose at 24 and 48 hours. When the patients were pretreated with potassium perchlorate, the uptakes were approximately 12 percent and 21 percent of the administered dose at 24 and 48 hours following the administration of the tracer.

The duration of the inhibition of iodide uptake after the oral administration of 100 mg of potassium perchlorate (71.8 mg of perchlorate) appeared to be about six hours. Beyond six hours, accumulation of I^{131} commenced. Durand (1938; as cited in Stanbury and Wyngaarden, 1952) found that at this time approximately half the administered dose of perchlorate has been excreted in the urine.

A LOAEL is not identified for this experiment because: (a) there were only two subjects that completed the experiment, (b) acute exposure, and (c) the patients suffered from a thyroid disease which might have affected iodide uptake by the thyroid.

Godley and Stanbury (1954) reported using potassium perchlorate to treat 24 patients with Graves' disease. Patients were treated with 600 to 1,200 mg/day for at least 11 weeks with a few as long as 52 weeks. Thirteen patients had determinations of the uptake of radioactive iodine by the thyroid both before beginning perchlorate therapy and

within two weeks after medication had begun. The mean control uptake was 77.5 percent, with a range from 60.7 to 108 percent. The mean uptake during perchlorate therapy was 15.9 percent, with a range from 3.4 to 38.8 percent.

Bürgi *et al.* (1974) studied the effects of perchlorate treatment on the release of endogenous iodine from the thyroid glands of five normal healthy volunteers (three females and two males). The volunteers were given ^{125}I -labeled iodide and ^{131}I -labeled T4 for seventeen days followed by 3 x 200 mg/day (9.7 mg/kg-day) perchlorate for eight days. Analysis for the two tracers in the urine and serum of the subjects showed that this dose was sufficient to totally block iodide uptake by the thyroid. Additionally, the perchlorate treatment caused an increase in excretion of non-thyroxine iodine of 65 percent above background.

Other Human Studies

Brabant *et al.* (1992) studied five healthy male volunteers pretreated with 200 µg/day iodine for four weeks before perchlorate exposure. Iodine exposure was discontinued, and the volunteers were given 3 x 300 mg/day of perchlorate for another four weeks. Serum levels of T3 and T4 were measured at the end of the four-week perchlorate-dosing period. Perchlorate treatment had no effect on serum T3 or T4 levels or on thyroid gland volume. However, serum free T4 and TSH levels were significantly diminished by treatment, and thyroglobulin serum levels were almost doubled, indicating the stress of the treatment on the thyroid hormone balance. It is interesting to note that Brabant *et al.* (1992) found the perchlorate treatment significantly reduced intrathyroidal iodine concentration levels.

In a follow-up study, Brabant *et al.* (1994 as cited in U.S. EPA, 2002) repeated the earlier studies with perchlorate treatment lasting longer than 4 weeks. As a result of the longer treatment, thyroid volumes increased in all subjects although TSH levels did not increase.

Lawrence *et al.* (2000) administered perchlorate to nine healthy male volunteers and monitored the impact on thyroid function. Each subject ingested 10 mg of perchlorate (as potassium perchlorate) dissolved in a liter of spring water during waking hours for 14 days. Baseline serum TSH, total T4, total T3, 24-hour thyroid I^{123} uptakes, serum and 24-hour urine perchlorate, and 24-hour urine iodine were determined. All blood and urine tests were repeated on days 7 and 14 of perchlorate administration and 24-hour thyroid I^{123} uptakes on day 14 of perchlorate administration. All tests were repeated 14 days after perchlorate exposure was discontinued. No effect of perchlorate was observed on serum T4, T3 and TSH (Table 10). It should be noted that the dietary iodine intake levels of the subjects were rather high as indicated by the high urine iodine values (Table 11). Because iodide and perchlorate compete for the same receptor site on the sodium-iodide symporter (Wolff, 1998), a high dietary iodide intake may reduce the impact of perchlorate on the thyroid. It is also interesting that there was no statistical difference in serum perchlorate levels after 7 days and 14 days of exposure, indicating no apparent accumulation of perchlorate in the systemic circulation over that period of time.

Table 10. The Effect of Perchlorate Administration (10 mg/day, about 0.14 mg/kg-day) for 7 and 14 Days on Thyroid Function Tests (Lawrence *et al.*, 2000)

Time	T ₄ (µg/dL)	T ₃ (ng/dL)	TSH (µU/mL)
Baseline	6.6±0.4	136±6	1.05±0.14
Day 7 during perchlorate ingestion	6.7±0.4	140±8	1.00±0.17
Day 14 during perchlorate ingestion	6.6±0.5	151±6	0.96±0.12
14 days after perchlorate was discontinued	6.5±0.5	157±9	1.23±0.17

Values are mean±standard error

Table 11. Urine and Serum Perchlorate and Iodine Values Before, During, and After Ingestion of 10 mg Perchlorate (about 0.14 mg/kg-day) for 14 Days (Lawrence *et al.*, 2000)

Time	Urine perchlorate (mg/24 hr)	Serum perchlorate (µg/ml)	Urine iodine (µg/24 hr)	Serum iodine (µg/dL)
Baseline	<0.5	0	254±69	6.5±0.42
Day 7 during perchlorate ingestion	7.7±0.8	0.61±0.02	233±49	6.2±0.34
Day 14 during perchlorate ingestion	7.5±1.0	0.59±0.02	385±123	6.4±0.37
14 days after perchlorate was discontinued	<0.5	0	208±42	6.3±0.57

Values are mean±standard error

Lawrence *et al.* (2000) also reported that during perchlorate ingestion, there was a highly significant decrease in the thyroid I¹²³ uptakes at all three time points. In each instance, 150 µCi I¹²³ was administered to a subject and thyroid iodide uptake was measured at 4, 8, and 24 hours. They reported that the average decrease below baseline values over all three time points was 38 percent. Two weeks after perchlorate was discontinued, the 24-hour thyroid I¹²³ uptakes were significantly higher than baseline at 4, 8, and 24 hours (Table 12).

Table 12. Thyroid I¹²³ Uptakes Before, During, and After Ingestion of 10 mg Perchlorate (about 0.14 mg/kg-day) Daily for 14 Days (Lawrence *et al.*, 2000)

Time	Thyroid I ¹²³ uptake -baseline (% of dose)	Thyroid I ¹²³ uptake 14 days on perchlorate (% of dose)	Thyroid I ¹²³ uptake 14 days after perchlorate was discontinued (% of dose)
4 hours	12.5 ± 1.3	8.2 ± 0.7 ^a	16.6 ± 2.4 ^b
8 hours	17.3 ± 1.9	10.6 ± 1.0 ^a	21.9 ± 2.8 ^b
24 hours	23.6 ± 2.6	14.0 ± 1.6 ^a	27.1 ± 3.3 ^c

^a p < 0.01 vs. baseline and after perchlorate treatment was discontinued

^b p < 0.01 vs. baseline

^c p < 0.05 vs. baseline

In a follow-up study, Lawrence *et al.* (2001) administered a daily oral dose of 3 mg to a group of eight healthy male volunteers for 14 days. They reported that the mean 8-hr thyroid radioactive iodine uptake decreased from 13.1 percent to 11.8 percent, during perchlorate ingestion. Similarly, the 24-hr thyroid radioactive iodine uptake decreased from 16.1 percent to 14.5 percent, a small but statistically insignificant 10 percent decrease. Details of the study were not reported. Assuming a default body weight of 70 kg, the dose used in this study is estimated to be 0.043 mg/kg-day.

Greer *et al.* (2002) described two studies in which daily oral doses of perchlorate (ClO₄) dissolved in 400 ml water were given to groups of euthyroid human volunteers for 14 days. Subjects (4 male and 4 female; 18-57 years old) of each dose group were exposed to a daily dose of 0.02, 0.1, or 0.5 mg/kg perchlorate (approximately 1.4, 7, or 35 mg, assuming 70 kg body weight). In a follow-up study, one additional subject of each sex received perchlorate at 0.02, 0.1, or 0.5 mg/kg-day, while six women and one man received a dose of 0.007 mg/kg-day. Subjects drank 100 ml of the perchlorate solution at 4 set times throughout each day. Measurement of 8- and 24-hour I¹²³ thyroid uptakes was performed prior to perchlorate exposure (baseline), on exposure days 2 and 14, and on post-exposure day 15. There was a strong correlation between the 8- and 24-hr uptakes over all dose groups and measurement days. There was no difference between exposure days 2 and 14 in the inhibition of uptake produced by a given perchlorate dose. There was no sex difference. Uptakes measured on post-exposure day 15 were not significantly different from baseline. Table 13 provides the 24-hr thyroid radioiodine uptake data by dose. Greer *et al.* (2002) measured total T4, free T4, total T3, and TSH in blood sampled throughout the study, and found them to be in the normal range for all subjects. One woman in the lowest dose group had abnormally high TSH on both the screening visit and on exposure day 14. Because of the limited data in the follow-up study, analysis of treatment effects on hormonal levels was confined to the 24 volunteers in the first study. Greer *et al.* (2002) reported no association of thyroid hormone levels with blood-draw events in any dose group except a marginally significant association of TSH with blood-draw in the 0.5 mg/kg-day dose group.

Table 13. Descriptive Statistics for the 24-hr Thyroid Radioiodine Uptake Data (from Greer *et al.*, 2002)

Dose group	Time	Number of subjects in dose group	24-hr uptake (mean±standard error)	
			Raw (%)	% of baseline
0.007 mg/kg-day	Baseline visit	7	18.1±3.1	
0.007 mg/kg-day	Exposure day 14	7	16.5±1.6	98.2±8.3
0.02 mg/kg-day	Baseline visit	10	18.4±1.2	
0.02 mg/kg-day	Exposure day 14	10	15.2±1.1	83.6±4.1
0.1 mg/kg-day	Baseline visit	10	19.9±2.1	
0.1 mg/kg-day	Exposure day 14	10	11.0±1.6	55.3±3.9
0.5 mg/kg-day	Baseline visit	10	21.6±2.0	
0.5 mg/kg-day	Exposure day 14	10	6.9±0.9	32.9±3.8

Occupational Studies

Gibbs *et al.* (1998) monitored triiodothyronine resin uptake (T3U), total serum T4, free T4 index (FTI) and serum TSH levels of 18 workers occupationally exposed to ammonium perchlorate in air before and after a work shift. They also similarly monitored 83 workers who were not exposed. Based on the thyroid function test results collected, Gibbs *et al.* (1998) concluded that exposure to a mean of 36 µg/kg-day ammonium perchlorate (ranging from 0.2 to 436 µg/kg-day) was not a significant predictor of the cross-shift change in any of the thyroid parameters. Given the relatively long serum half-life of T4 in humans (5-9 days) (U.S. EPA, 1998b), it would be very unlikely that T4 would exhibit a change in serum over a single work shift.

Gibbs *et al.* (1998) also evaluated the thyroid function test results of workers exposed to ammonium perchlorate based on their working-lifetime dose estimates. They reviewed personnel records and employees were interviewed to determine the number of years worked in each of the seven exposure groups. An average of 2,000 hours worked yearly was assumed. Each subject's working-lifetime cumulative dose was then estimated as:

$$\Sigma[\text{mean group exposure}] \times [\text{years in exposure group}] \times 2,000$$

No significant correlations with estimated lifetime cumulative perchlorate dose were detected with any of the measures of thyroid function (T3U, T4, FTI, and TSH levels). However, the tenure of the workers ranged from 1 to 27 years (Gibbs *et al.*, 1998), while thyroid hormone levels are most likely to be affected by relatively recent perchlorate exposures (probably in the range of 1-3 months). Cumulative dose over a long period of time may not be the best metric for characterizing the effect of perchlorate exposure on thyroid hormone levels.

Lamm *et al.* (1999) conducted a cross-sectional study of two similar worker populations from the same industrial complex: ammonium perchlorate production workers and sodium azide production workers. A total of 37 workers were exposed to airborne ammonium perchlorate, 35 males and two females. Twenty-one workers from the azide production plant served as the control group. Perchlorate exposure was measured using full-shift breathing zone personal air samplers for total as well as respirable perchlorate particles. Urinary perchlorate concentration was assessed at the beginning and end of the 12-hour shift in which the perchlorate exposure was measured. Lamm *et al.* (1999) reported that there were no differences in thyroid function tests between workers in the azide and perchlorate plants or between the azide workers and any of the three perchlorate-exposure groups (Table 14). Based on these data, a NOAEL of 0.48 mg/kg-day (33.6 mg/day divided by 70 kg) can be estimated. However, it should be noted that this data set has several limitations: (a) small sample size, (b) high dietary iodine intake among the workers, and (c) given the short biological half-life of perchlorate (approximately 8 hr), the exposed workers might recover from the effects of perchlorate during off-shift hours. Using the medical examination and questionnaire findings, Lamm *et al.* (1999) reported that worker exposures to perchlorate in the plant were not found to be associated with thyroid abnormalities.

Hematological Effects

Graves' disease patients treated with perchlorate doses in the range of 6 to 14 mg/kg-day for three to eight months occasionally developed fatal aplastic anemia (Fawcett and Clarke, 1961; Hobson, 1961; Johnson and Moore, 1961). Mechanism of this blood disorder is not known. The use of perchlorate to treat Grave's disease was discontinued because of these cases. Nonfatal agranulocytosis was reported in patients treated with 14 mg/kg-day perchlorate for 12 days (Southwell and Randall, 1960) or three months (Sunar, 1963). Barzilai and Sheinfeld (1966) reported that 8 of 76 patients treated with 14 mg/kg-day perchlorate for at least two months developed leukopenia or other side effects. There was also one case of fatal aplastic anemia and one of fatal agranulocytosis within this group of 76 patients (Barzilai and Sheinfeld, 1966). As similar adverse hematological effects were not observed in rodents exposed to 30 mg/kg-day or 100 mg/kg-day, humans may be more sensitive than rodents for this endpoint.

Table 14. Perchlorate Exposures and Thyroid-Function Parameters, by Plant and Exposure Groups (Adapted from Lamm *et al.*, 1999)

Group	Total airborne perchlorate exposure (mg/day)	Respirable airborne perchlorate exposure (mg/day)	Absorbed dose, derived from urinary perchlorate levels (mg/shift)	T4 (µg/dL)	T3 (ng/dL)	TSH (µU/mL)
Normal range				5 to 11	87 to 178	0.45 to 4.5
Azide worker	0.014±0.012 (n=4)	0.021±0.014 (n=6)	0.88±1.17 (n=21)	6.73±1.48 (n=21)	142.5±17.5 (n=21)	3.14±1.87 (n=21)
Perchlorate worker A	0.337±0.187 (n=6)	0.091±0.095 (n=11)	3.98±2.69 (n=14)	7.13±1.58 (n=13)	148.4±25.2 (n=13)	2.68±1.14 (n=12)
Perchlorate worker B	6.57±7.14 (n=2)	0.601±0.671 (n=7)	10.9±8.7 (n=8)	7.34±1.12 (n=8)	152.1±23.2 (n=8)	2.41±1.27 (n=8)
Perchlorate worker C	59.4±53.6 (n=12)	8.59±9.39 (n=14)	33.6±14.5 (n=14)	7.03±1.30 (n=15)	152.1±20.4 (n=15)	3.33±2.34 (n=15)

Values are mean±standard deviation
n = sample number

Carcinogenicity

Li *et al.* (2001) studied the prevalence of eight thyroid diseases (including malignant neoplasm of the thyroid gland) in Clark County, Nevada compared to that in Washoe County, Nevada and in other Nevada counties using Nevada Medicaid data for 1997 and 1998. Eighty-six measurements were taken of the perchlorate levels in the water supply of Clark County (Las Vegas) between July 1997 and January 2001. These values ranged from non-detect to 24 µg/L (ppb), with a mean of 11.5 µg/L. The researchers reported that none of the eight thyroid disease prevalences was significantly higher in Clark County than in the rest of the state, and for acquired hypothyroidism the prevalence in Clark County was significantly lower in the rest of the state. They also found none of the eight thyroid diseases significantly differed in prevalence between Clark County and Washoe County, with all relative risks being close to one.

Morgan and Cassady (2002) assessed observed and expected number of new invasive cancer cases for all sites combined and 16 cancer types among residents of the greater Redlands area between 1988 and 1998. The community is known to have drinking water contaminated with perchlorate (levels as high as 98 ppb measured in 2001) and trichloroethylene (0.09-97 ppb measured in 1980). They reported no significant differences between observed and expected numbers for all cancers, thyroid cancer, or 11 other cancer types. Significantly fewer cases were observed than expected for cancer of the lung and bronchus and the colon and rectum. More cases were observed for uterine cancer (standardized incidence ratio = 1.35; 99 percent CI = 1.06 to 1.70) and skin melanoma (standardized incidence ratio = 1.42; 99 percent CI = 1.13 to 1.77).

Adverse Health Effects Associated with Iodide Deficiency or Low Thyroid Hormone Levels

The most important and early effect of perchlorate exposure is its reduction of iodide uptake by the thyroid. For this reason, adverse health effects associated with iodide deficiency are discussed in this section.

Thyroid problems in pregnant women with low iodide intake

A number of human studies have shown that pregnancy stresses the thyroid (Crooks *et al.*, 1967; Glinioer *et al.*, 1990, 1992, 1995; Smyth *et al.*, 1997; Caron *et al.*, 1997; Brent, 1999; Kung *et al.*, 2000). In areas of iodine deficiency (intake level <100 µg/day), there is an increased risk of abnormally low serum T3 and T4 levels, thyroid enlargement as well as goiter in pregnant women. The nature and severity of changes in thyroid functions are related to the severity of iodine deficiency. In an epidemiologic survey reported by Delange and Ermans (1991; as cited in Delange, 1994), the investigators found the prevalence of goiter in an area with severe iodine deficiency is influenced by age and sex, with maximal frequency in females during puberty and childbearing age (Figure 5).

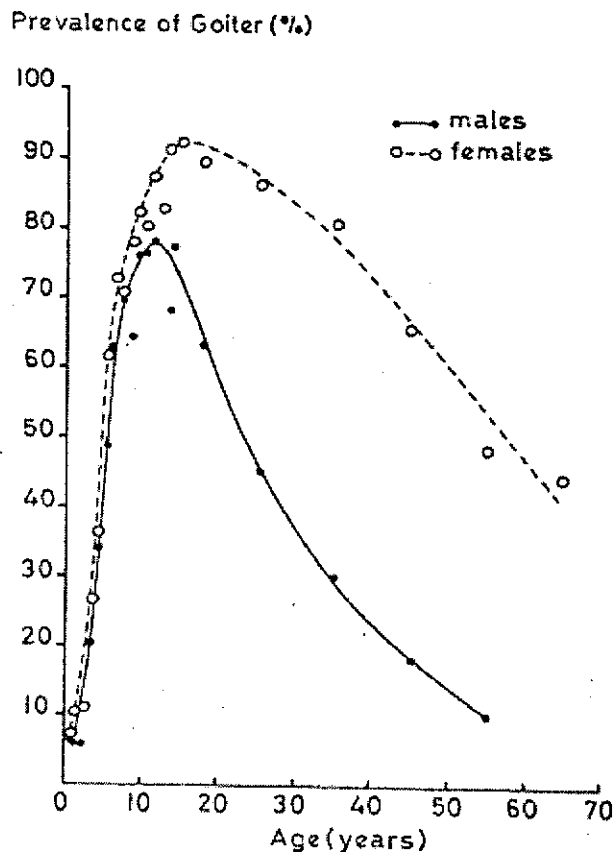


Figure 5. Changes in the prevalence of goiter as a function of age and sex in severe endemic goiter (Idjwi Island, Zaire) (from Delange, 1994).

Crooks *et al.* (1967) studied enlargement of the thyroid gland in pregnant and non-pregnant women in Aberdeen, Scotland and Reykjavik, Iceland. In the Scotland study, they found that the thyroid gland was visible and palpable in 70 percent of pregnant women but in only 37 percent of non-pregnant women in the reproductive age group. By contrast, in the Iceland study, the frequency of thyroid enlargement was about the same in pregnant (23 percent) as in non-pregnant women (19 percent). Crooks *et al.* (1967) suggested that the results can be explained by the fact that Icelandic diet is based on fish and contains a lot of iodine. This hypothesis is supported by the significantly higher mean plasma-inorganic-iodine concentration measured in non-pregnant Icelandic women ($0.691 \mu\text{g/dL}$) compared to the mean of $0.420 \mu\text{g/dL}$ found in Scottish non-pregnant women ($p < 0.001$).

Glinoe *et al.* (1990) suggested that in conditions of marginally low iodide intake, pregnancy constituted a goitrogenic stimulus. They followed prospectively a group of 606 healthy pregnant women in Brussels, Belgium, an area of marginally low iodide intake ($50\text{--}70 \mu\text{g/day}$), and monitored their T3, T4, TSH, and human chorionic

gonadotropin (hCG) levels in serum during the first, second, and third trimesters. All subjects were evaluated clinically and determined to be without detectable thyroid abnormality at the beginning of the study. Glinoe *et al.* (1990) found that a normal thyroid is faced with a triple challenge during pregnancy. First, there is a significant increase in circulating levels of the major T4 transport protein, thyroglobulin, in response to high estrogen levels. As a result, the thyroid has to increase its T4 output in order to maintain a stable T4/thyroglobulin ratio of 37-40 percent.

Second, several thyroidal stimulating factors of placental origin (mainly hCG) are produced in excess. This contributes to a decrease of serum TSH (mainly in the first half of gestation) and an increase in thyroid volume (Table 15). They found that the average thyroid volume increased by 18 percent of the initial size determined at the beginning of the pregnancy. The increase was significant and thyroid size increased in a majority of women (73 percent). Goiter, defined as thyroid volume greater than 23 mL, was found in 9 percent of the cohort at delivery.

Table 15. Changes in Mean Thyroid Volume in Healthy Women During Pregnancy (from Glinoe *et al.*, 1990)

Stage of Pregnancy	n	Total volume (mL)
First trimester	168	12.1±4.5
Second trimester	172	12.8±4.5
Third trimester	33	13.9±4.8 *
Delivery	179	15.0±6.8 **

* $p < 0.03$

** $p < 0.001$

Third, pregnancy is accompanied by a decrease in the availability of iodide for the maternal thyroid, due to increased renal clearance (Aboul-Khair *et al.*, 1964; as cited in Glinoe *et al.*, 1990) and losses to the feto-placental complex during late gestation, resulting in a relative iodine deficiency state.

In a related study, Glinoe *et al.* (1992) monitored thyroid condition of pregnant women in an area without overt iodine deficiency, but with a marginal iodide supply (less than 100 µg/day in 80 percent of women). They found that maternal thyroid function was characterized at delivery by relative hypothyroxinemia; increased T3/T4 ratios, indicating preferential T3 secretion; slightly increased TSH levels within the normal range in 97 percent of women; increased serum thyroglobulin values, which were above normal in 60 percent of women; and also goiter formation in almost 10 percent of women. In the newborns, they found free T4 levels were significantly higher than in the respective mothers. However, mean neonatal TSH and Tg levels were significantly higher than maternal values. Furthermore, these values were highly correlated with maternal data, suggesting the limited availability of iodine supply was the common link.

In a review paper, Glinoe (2001) again stressed the profound alterations in the thyroid economy associated with pregnancy. In healthy iodine-sufficient pregnant women, this leads to a physiological adaptation of the thyroid and an increased production of thyroid hormones. When gestation takes place in conditions with iodine restriction or deficiency, pregnancy may lead to pathological alterations affecting both thyroid function and the anatomical integrity of the thyroid gland. The more severe the iodine deficiency, the more obvious, frequent, and profound the potential maternal and fetal repercussions.

Supportive results were reported in two other studies, one in Ireland and the other in France. Smyth *et al.* (1997) evaluated ultrasound-measured thyroid volume of 115 pregnant women during one of the three trimesters. These women (Group A) were enrolled based on availability, and each trimester's study group comprised different individuals. Control values for thyroid volume were obtained from 95 pre-menopausal females. All subjects were from Dublin, Ireland, an area of moderate dietary iodide intake (median urinary iodine was 82 µg/day). All pregnant women studied delivered live-born, normally formed, singleton infants and received no iodide-containing supplements during their pregnancy. Smyth *et al.* (1997) reported that the mean thyroid volume of 13.9±0.8 mL, observed in the first trimester, was significantly greater than the control value (11.3±0.5 mL; $p < 0.05$) and reached a maximum of 16.0±0.7 mL, a 47 percent increase ($p < 0.01$), in the third trimester.

In a related study, Smyth *et al.* (1997) studied a group of 38 pregnant women (Group B), prospectively. Casual urine samples were collected sequentially during the 3 trimesters of pregnancy and at approximately 6 weeks postpartum. Of those 38 subjects, 20 had thyroid ultrasound scans during each trimester of pregnancy and at 6 weeks postpartum. Thyroid volumes greater than 18.0 mL were defined as enlarged. The number of enlarged thyroids increased from the non-pregnant control value of 6.3 percent, through 19.5 percent in the first trimester, to reach a plateau of approximately 32 percent in the second and the third trimesters, which was maintained up to 40 days postpartum.

Urinary iodine of the women in Group A and Group B were also measured. Urinary iodine measurements from 1063 premenopausal women over a one-year period served as controls. Smyth *et al.* (1997) found that urinary iodine levels measured throughout the pregnancies of the women in Group A and Group B (Table 16) were higher than in the controls (median 70 µg/L). Smyth *et al.* (1997) suggested that in an area of moderate dietary iodide intake, urinary loss during pregnancy may result in maternal thyroid enlargement.

Table 16. Median Urinary Iodine Excretion (µg/L) in Pregnancy (from Smyth *et al.*, 1997) *

	First Trimester	Second Trimester	Third trimester
Group A	135	122	122
Group B	155	122	115

* Some of the values were estimated from a graph.

In a prospective study, Caron *et al.* (1997) evaluated thyroid condition of 347 pregnant women living in the southwest of France (with an estimated urinary iodine excretion value of 50 µg/day). Iodine concentration in urine samples and serum thyroid hormone measurements were taken at initial presentation (before 12 weeks of gestation), and during the nine months of pregnancy. Mean urinary iodine levels were low during the first trimester (6.9 ± 0.4 µg/dL), as well as during the ninth month of pregnancy (8.6 ± 0.6 µg/dL). A thyroid ultrasound was performed one to five days after delivery in 246 mothers. During pregnancy free T4 and T3 concentrations decreased ($p < 0.001$), and TSH and Tg concentrations increased ($p < 0.001$). Thyroid hypertrophy (thyroid volume greater than 18 mL) was present in 29 percent of the mothers. The percentage of thyroid hypertrophy at delivery was associated with urinary iodine concentration during the first trimester of gestation, 15.4 percent (urinary iodine < 5 µg/dL), 9.2 percent (urinary iodine 5-10 µg/dL), and 3.5 percent (more than 10 µg/dL) urinary iodine (Figure 6). Goiter (thyroid volume greater than 22 mL) was present in 11 percent of the mothers. They concluded that in areas with a marginally low iodide supply, pregnancy constitutes a goitrogenic stimulus.

In another prospective study, Kung *et al.* (2000) studied 230 pregnant women living in a borderline iodine sufficient area (Hong Kong). The median urine iodine concentration in healthy adults was 0.77 µmol/L (9.8 µg/dL) in Hong Kong, which was close to the World Health Organization cut-off value of 0.79 µmol/L (or 10 µg/dL) for iodine sufficiency. When recruited into the study, all pregnant women were in their first trimester; subjects with a history of thyroid dysfunction were excluded. These women were prospectively studied at approximately 12-14 weeks, 20-24 weeks, and 36 weeks of gestation, as well as 6 weeks and 3 months postpartum for thyroid function, thyroid volume by ultrasound examination, and urine iodine concentration. Study results are presented in Table 17. The investigators showed that in an area of borderline low dietary iodine intake, pregnancy was an important stress to the maternal thyroid axis. Pregnancy caused an average of 30 percent increase (range 3 percent – 230 percent) in thyroid volume, with some subjects having more than twofold increase in their thyroid volume. This thyroid enlargement persisted and failed to revert completely even 3 months after delivery.

They also reported that 14 women with excessive thyroidal stimulation in the second trimester had lower urine iodine concentrations and larger thyroid volumes throughout pregnancy. Furthermore, their neonates had higher cord TSH, Tg, and slightly higher thyroid volumes as compared to the findings in 216 pregnant women without evidence of thyroid stimulation. Seven neonates (50 percent) born to these women had subnormal free T4 levels at birth.

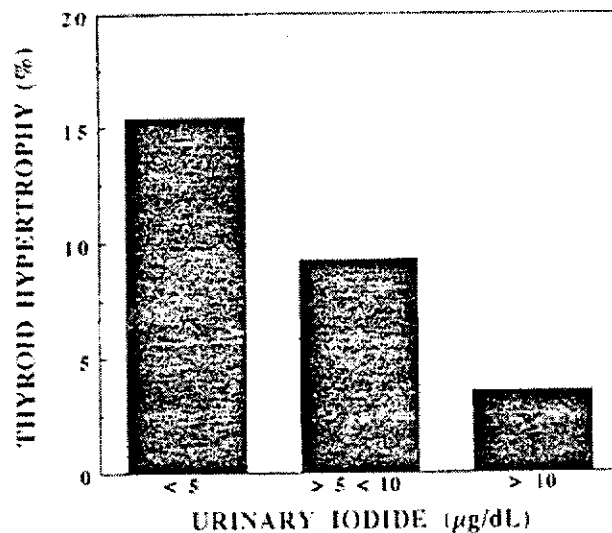


Figure 6. Percentage of maternal thyroid hypertrophy (thyroid volume > 18 ml) in relation to urinary iodine concentration during the first trimester of pregnancy (from Caron *et al.*, 1997).

Table 17. Change of Thyroid Function Tests, Thyroidal Volume, and Urinary Iodine Level of Women During and After Pregnancy (Kung *et al.*, 2000)

	First trimester	Second trimester	Third trimester	Postpartum 6 weeks	Postpartum 3 months
Total T4 (nmol/L)	154 (132-176)	126 (110-143) *	125 (106-142) *	89 (81-98) §	92 (82-101) §
Free T3 (pmol/L)	3.9 (3.6-4.3)	3.4 (3.1-3.7) *	3.3 (3.0-3.7) *	4.0 (3.7-4.4)	4.3 (4.1-4.6)
Free T4 (pmol/L)	13.4 (12.2-15.0)	11.9 (10.7-13.1) *	11.7 (10.1-13.0) *	14.5 (13.1-16.0)	14.4 (13.0-15.8)
TSH (mIU/L)	0.49 (0.12-1.00)	0.96 (0.62-1.28) *	0.95 (0.60-1.36) §	1.15 (0.74-1.58) §	1.14 (0.81-1.61) §
Urine iodine (µmol/L)	0.84 (0.60-1.09)	0.91 (0.65-1.14) *	0.98 (0.72-1.24) *	0.83 (0.56-1.08)	0.79 (0.51-1.14)
Thyroid volume (ml)	9.5 (7.2-12.3)	10.3 (7.7-13.6) *	11.2 (8.9-13.8) *	11.0 (8.3-14.2) *	10.6 (8.6-13.7) *

Results are median, * $p < 0.05$, § $p < 0.01$, vs. first trimester

Another source of data supporting the concept that normal pregnancy requires increased thyroid hormone production comes from the observation that women previously diagnosed with hypothyroidism on adequate T4 replacement doses often require an increase in their T4 doses during pregnancy (Table 18).

Table 18. Thyroid Hormone Requirement in Pregnancy (from Brent, 1999)

Study	Mean daily dose (μg)	Fraction of women requiring an increased dose	Mean dose increase for those who had an adjustment (μg)
Pekonen <i>et al.</i> (1984)	141	7/34 (21%)	62
Mandel <i>et al.</i> (1990)	148	9/12 (75%)	46
Tamaki <i>et al.</i> (1990)	-	4/4 (100%)	-
Girling and de Swiet (1992)	142	9/32 (28%)	68
Kaplan (1992)	154	27/42 (64%)	42
Pooled data	146	56/124 (45%)	46

There are three prospective studies showing that in an area with marginal or moderate iodide deficiency, iodide supplement often can reduce the stress on the thyroid during pregnancy (Romano *et al.*, 1991; Pedersen *et al.*, 1993; Glinoer *et al.*, 1995). The first study was carried out in L'Aquila, Italy, an area with moderate iodine deficiency (Romano *et al.*, 1991). There were 35 pregnant women in the study, all of them had a normal pregnancy and no personal history of thyroid disease. They had a mean age of 27.1 ± 3.8 years and a mean body weight of 61.6 ± 4.9 kg at the first examination during the first trimester. Pregnant women were randomly assigned into group A (n=17) or group B (n=18). Immediately after the first examination, iodide salt equivalent to a daily intake of about 120 to 180 μg iodide was prescribed to all the women in group A. Each trimester all pregnant women in both groups were subjected to three ultrasonographic evaluations of thyroid volume and to measurement of body weight. During each examination, 24-hour urine samples were also taken to determine the iodine urinary excretion. Romano *et al.* (1991) reported that TSH levels of all the subjects were within the normal range and TSH levels measured in group A did not statistically differ from those measured in group B. The effect of iodide supplement was confirmed by urinary iodine measurements. A significant increase in urinary iodine excretion was found at the second and third examination ($p < 0.0001$ and $p < 0.01$, respectively, Table 19) only in group A, treated with iodide salt.

Table 19. Iodine Excretion ($\mu\text{g}/24$ hours) in Both Groups at Each Trimester (mean \pm standard deviation) (from Romano *et al.*, 1991)

	First trimester	Second trimester	Third trimester
Group A	37.0 \pm 36.0	154.0 \pm 59.0 *	100.0 \pm 39.0 **
Group B	30.5 \pm 42.0	55.0 \pm 35.0	50.0 \pm 37.0

* $p < 0.0001$

** $p < 0.01$ vs. first trimester

Thyroid volume did not change significantly throughout pregnancy in the group treated with iodide salt, whereas in the control group (Group B) it increased significantly ($p < 0.0001$), with a mean increase of 1.6 ± 0.6 ml (16.25 percent \pm 6.29 percent) between the first and third trimester. Romano *et al.* (1991) concluded that an adequate dietary iodide intake is necessary to prevent the development of a gestational goiter, and iodine deficiency is the main causative cofactor of thyroid enlargement during pregnancy.

A similar study was also carried out in East Jutland, Denmark, an area with a median daily urinary iodine excretion around 50 μ g (Pedersen *et al.*, 1993). The researchers selected 54 normal pregnant women and randomly divided them into treated (28 subjects) and control groups (26 subjects). Before iodine supplementation was initiated, the measured variables were nearly identical in the two groups. Treated subjects received 200 μ g iodide/day starting from weeks 17-18 of pregnancy until 12 months after delivery. All women were followed at regular intervals during pregnancy. In the control group, serum TSH, serum Tg, and thyroid size showed significant increases during pregnancy. These variations were ameliorated by iodide supplementation (Figures 7, 8, and 9).

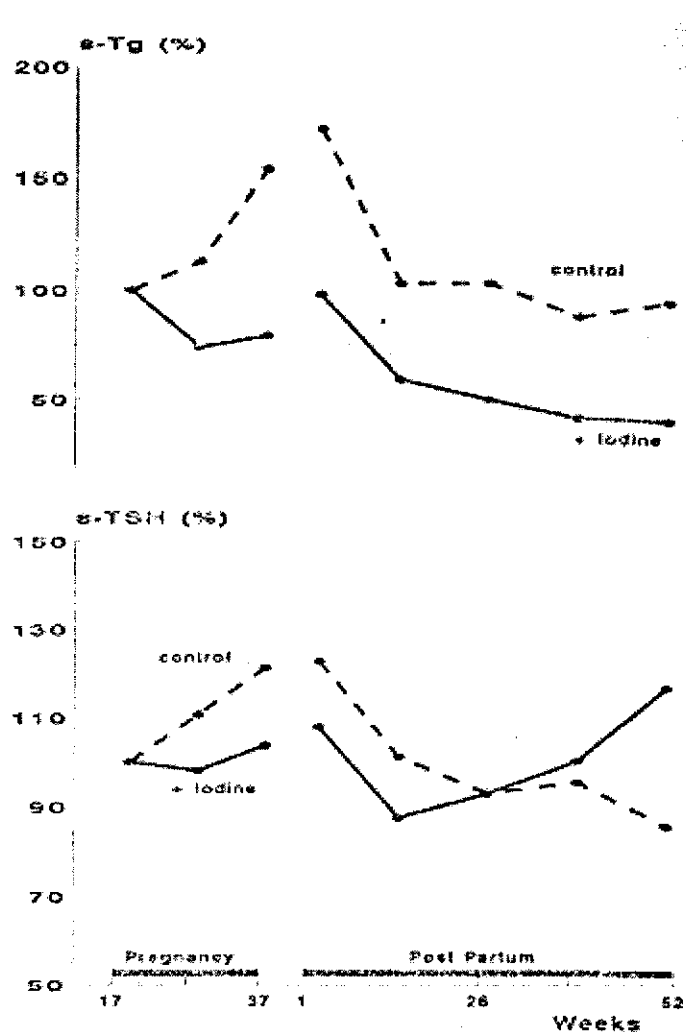


Figure 7. Serum Tg and TSH during pregnancy and for 52 weeks postpartum in women receiving iodide supplementation and control women, as a percentage of the initial values. Median values are shown. The increase in serum Tg during pregnancy in the control group was statistically significant ($p < 0.01$), but the first value obtained during pregnancy and the values obtained one year after delivery were not different. Tg values from the two groups were significantly different at all periods, except before initiation of iodide supplementation.

The increase in serum TSH in the control group during pregnancy was statistically significant ($p < 0.01$), whereas no differences between values were found in the iodide supplemented group ($p = 0.29$, by Friedman's test). During the postpartum period, no significant TSH differences between the groups were found (from Pedersen *et al.*, 1993).